

**MERIT SYSTEM SERVICES - SAMPLE ELIGIBILITY WORKER I - ITEMS****READING AND UNDERSTANDING WRITTEN MATERIALS**

**DIRECTIONS:** Read each question carefully. Select the best answer and darken the proper space on the answer sheet.

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1. Through advertising, manufacturers exercise a high degree of control over consumers' desires. However, the manufacturer assumes enormous risks in attempting to predict what consumers will want and in producing goods in quantity and distributing them in advance of final selection by the consumers.

The paragraph best supports the statement that manufacturers

1. distribute goods directly to the consumers
2. can eliminate the risk of overproduction by advertising
3. can predict with great accuracy the success of any product they put on the market
4. must depend upon the final consumers for the success of their undertakings

2. In the relations of humans to nature, the procuring of food and shelter is fundamental. With the migration of humans to various climates, ever new adjustments to the food supply and to the climate became necessary.

The paragraph best supports the statement that the means by which humans supply their material needs are

1. varied
2. limited
3. accidental
4. inadequate

3. "Some fire-resistant buildings, although wholly constructed of materials that will not burn, may be completely gutted by the spread of fire through their contents by way of hallways and other openings. They may even suffer serious structural damage by the collapse of metal beams and columns."

The quotation best supports the statement that some fire-resistant buildings

1. will burn readily
2. can be damaged seriously by fire
3. have specially constructed halls and doors
4. afford less protection to their contents than would ordinary buildings

# Eligibility Technician Exam Sample Questions

**Judith L. Neville**



## **Eligibility Technician Exam Sample Questions:**

**Mosby's Advanced Pharmacy Technician Exam Review-E-Book** James J. Mizner, 2023-12-21 From bestselling test preparation author James J Mizner comes Mosby's Advanced Pharmacy Technician Exam Review Available to Pharmacy Technicians with at least three years of work experience the Advanced Certified Pharmacy Technician CPhT Adv credential provides a pathway for obtaining higher level skills and advancing your career This new resource gives you the review and practice you need to prepare for the exam with an easy to use format sample certification exams content review chapters and more Make sure you're ready for exam and career success with this essential review Custom test generator on the Evolve companion website features practice and exam modes and timer functionality to strengthen topic expertise and simulate nearly unlimited unique exams for practice Sample certification exams in the book offer valuable test taking experience Dedicated chapter covering sterile compounding addresses an alternative pathway to certification as a Compounded Sterile Preparation Technician CSPT Content review chapters cover many advanced level certification topics Chapter review questions help reinforce knowledge and assess comprehension Many practice questions are available both in the print book and online for convenient access Bulleted listing format makes it easy to focus on reviewing the essentials **Resources in**

**Education**, 1992-04 **Pharmacy Technician Exam Review Guide** Judith L. Neville, 2013 Pharmacy Technician Exam Review Guide provides exam preparation help for pharmacy technician candidates preparing to sit for the Pharmacy Technician Certification Exam PTCE This book is designed to equip candidates with excellent study tools that are intended to review the required level of information to prepare the candidate for the examination by focusing on the key topic areas

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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton  $h_a + b \rightleftharpoons h_b$  strong acids e.g. sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the  $K_a$  or  $K_b$  from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $K_{a1}$  and  $K_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web jul 19 2023 at the beginning of the titration the solution in the erlenmeyer flask is acidic as the base is added it completely reacts with the acid and the solution in the erlenmeyer flask continues to be acidic but at the equivalence point

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the

neutral pH at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web 1 approximately 3 g of hydrated oxalic acid  $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$  was weighted to the nearest 0.01 g in a 50 mL beaker 2 about 30 mL of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 mL volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 mL the solution pH is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then  $\text{pH} = -\log 0.100 = 1.000$   $\text{pH} = -\log 0.$

100 1 000 b titrant volume 12 50 ml

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web one type of titration uses a neutralization reaction in which an acid and a base react to produce a salt and water in equation 1 the acid is hcl hydrochloric acid and the base is naoh sodium hydroxide when the acid and base react they form nacl sodium chloride which is also known as table salt

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