

## Chapter Assessment Chemistry: Matter and Change • Chapter 4

### Reviewing Vocabulary

Match each definition in Column A with the term in Column B.

#### Column A

#### Column B

- |  |                    |
|--|--------------------|
| 1. Radiation deflected toward the positively charged plate                         | a. atom            |
| 2. Atoms with the same number of protons but different numbers of neutrons         | b. nucleus         |
| 3. High-energy radiation that has no charge and no mass                            | c. atomic mass     |
| 4. The smallest particle of an element that retains the properties of that element | d.                 |
| 5. The weighted average mass of an element's isotopes                              | e. gamma ray       |
| 6. The center-most part of an atom where the protons and neutrons are contained    | f. alpha radiation |
| 7. Radiation deflected toward the negatively charged plate                         | g. beta            |
| 8. The rays and particles emitted by radioactive material                          | h. atomic          |
| 9. Equal to $1/12$ the mass of a carbon-12 atom                                    | i.                 |
| 10. Ray of radiation traveling from the cathode to the anode                       | j. Dalton's        |
| 11. Process not requiring energy by which unstable nuclei lose energy              | k. cathode         |
| 12. States that all matter is composed of atoms                                    | l.                 |
| 13. Process whereby some substances spontaneously emit radiation                   | m.                 |
- Compare and contrast each pair of related terms.
14. mass number, atomic number
15. nuclear reaction, nuclear equation

### Understanding Main Ideas

Use the periodic table to identify each element described below.

- atomic number 63
- 18 protons
- 44 protons and 44 electrons
- atomic number 24
- 21 protons
- atomic number 55

In the space at the left, write true if the statement is true; if the statement is false,

change the italicized term to make it true.

- An atom's nucleus contains its protons and electrons.
- Neutrons have no electrical charge.
- Beta particles have a charge of  $2^+$ .
- An alpha particle consists of two protons and two electrons.

Complete the table below.

# Glencoe Chemistry Matter And Change Chapter 2

## Assessment Answer Key

**R Sandford**



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