PRACTICAL ORGANIC CHEMISTRY

(A) FUNCTIONAL GROUPANALYSIS

- 1. Unsaturation : Alkenes & alkynes:
- (a) Bayers test: Cold dil alk. KMnO₄ decolourisation test Purple colour → Colourless + MnO₅ (Brown ppt)
- (b) Br, water decolourisation test
 Violet colourless of Br, → Colourless
- 2. Terminal alkynes:
 - Confirmed by ppt of Acetylide ion with NaNH, or AgNO, or Cu, Cl, NH, OH
- 3. Alkyl halides:
- (a) If they are capable of carbocation formation then they will give ppt with AgNO,
- (b) Beilstein's test: A green-colour is imported to the flame if small amount of organic compound is taken on copper wire.
- 4. Alcohol:
- (a) Cerric ammonium nitrate → Give red colour
- (b) Boil with acetic acid & conc. H₂SO₄ → fruity smell
- (c) 2-alkanol & ethanol also give Iodoform test → Yellow ppt. of CH, I on reaction with I,+ OH
- 5. Aldehyde & Ketones:
 - 2,4- Dinitrophenyl hydrazize (or) Braddy's reagent give yellow, orange or red color with ald. & Ketones (2,4-DNP)
- 6. Aldehydes:
- (a) Tollen's test → Silver mirror
- (b) Fehling's test {except benzaldehyde}→Red colour
- (c) Benedicts test → Red colour
- (d) Schiff's dye colour regeneration test → Pink colour
- (e) Gly ppt with HgCl₂.
- Ketones:
- (a) Methyl Ketones give haloform test
- (b) α-hydroxy Ketones give Tollen's & Fehling test's too.
- 8. Carboxylic acids:
- (a) Brisk effervescence with aq. NaHCO₃ solution.
- (b) HCOOH alone gives silver mirror test with Tollen's reagent.
- (c) Blue litmus → red
- (d) Give fruity smell on reaction with alcohols.
- 9. Phenols:
- (a) Violet colouration with neural FeCl,
- (b) Liebermann test
- (c) White ppt with Br, water
- (d) Brisk effervescence with aq. NaHCO₁ is observed in case of Nitrophenols.
- 10. Primary amines:
- (a) Carbylamine reaction → Isonitriles have very distinctive foul odors
- (b) Hoffmann mustard oil reaction → Oily liquid with mustard like smell.
- Aromatic 1° amine → diazo test

Give Me D Answer Of Chemistry Pratical Alternative A

Daniel F McAuley

Give Me D Answer Of Chemistry Pratical Alternative \boldsymbol{A} :

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