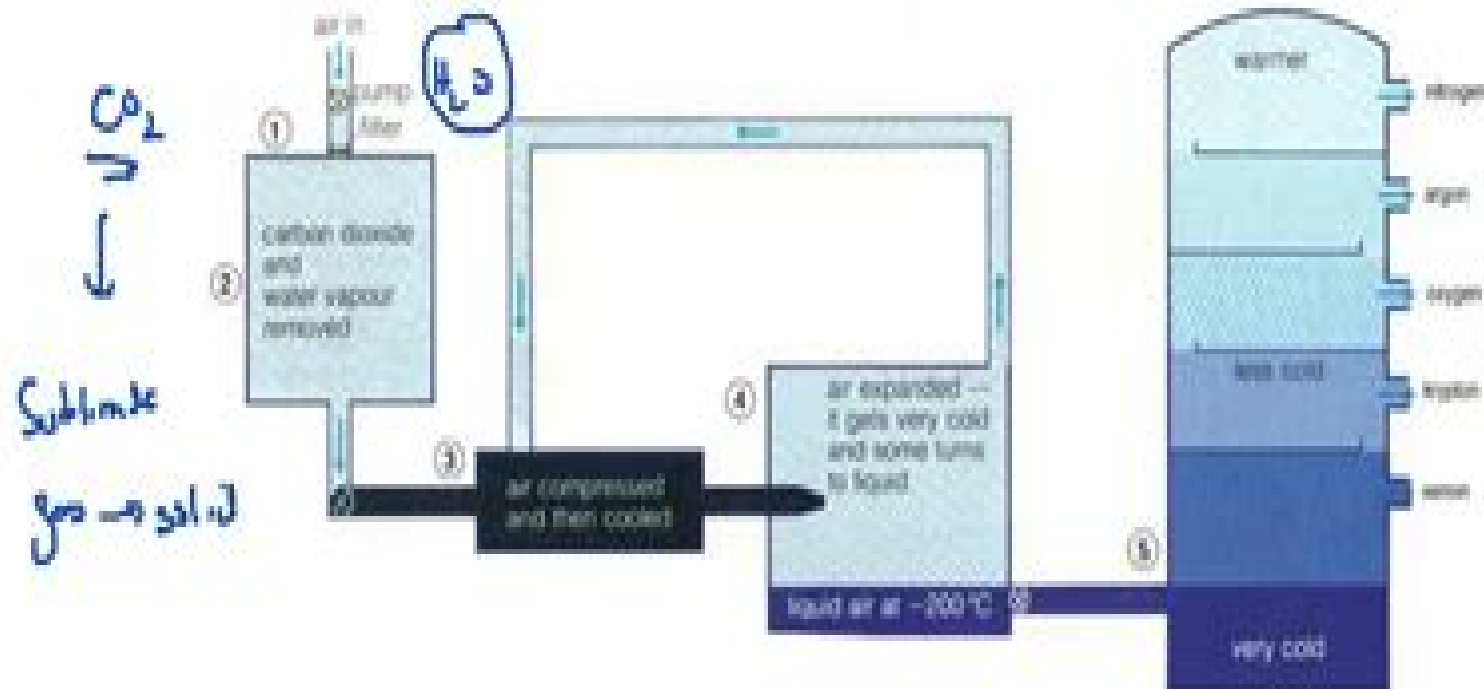


(Extended only) Describe the separation of oxygen and nitrogen from liquid air by fractional distillation

- Fractional distillation of liquid air separates oxygen and nitrogen
- oxygen has a boiling point higher than that of nitrogen
- air is liquefied (around -200°C)
- fractional distillation column temperature is increased until nitrogen boils and rises to the top as a gas (around -196°C)
- the nitrogen is collected and removed
- the temperature is increased until oxygen boils and rises to the top as a gas (around -183°C) and oxygen is collected



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Graham Hill



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Science to GCSE Stephen Pople, Michael Williams, 2002 Updated for the 2001 specifications Science to GCSE aims to

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Chemistry John Stranger Holman, Philip Stone, 2002 This science series had a curriculum audit matching the books to all the major specifications It has practical experiments expanded from the texts to include ICT support OHTs of all the diagrams in the textbooks are included Answers are given to all the questions in the textbooks Sc1 enquiry material is provided in line with the revised National Curriculum requirements It has additional support for Key Skills and additional material linked to the four learning programmes Science in Focus

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GCSE AQA Science Foundation Success Revision Guide, 2006-08 This Success Revision Guide offers accessible content to help students manage their revision and prepare for the exam efficiently The content is broken into manageable sections and advice is offered to help build students confidence Exam tips and techniques are provided to support students throughout the revision process

Folens GCSE Applied Science Colin Bell, 2003

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Chemistry at a Glance Roger Owen, Sue King, 2005-06-27 This book aims to cover the specifications of the main examination boards for GCSE Double Science GCSE Single Science and the core content of GCSE Chemistry Where relevant Key Stage 3 material is summarized as an introduction to GCSE topics This serves as revision of work done prior to Key Stage 4 and a foundation for GCSE studies The book is

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