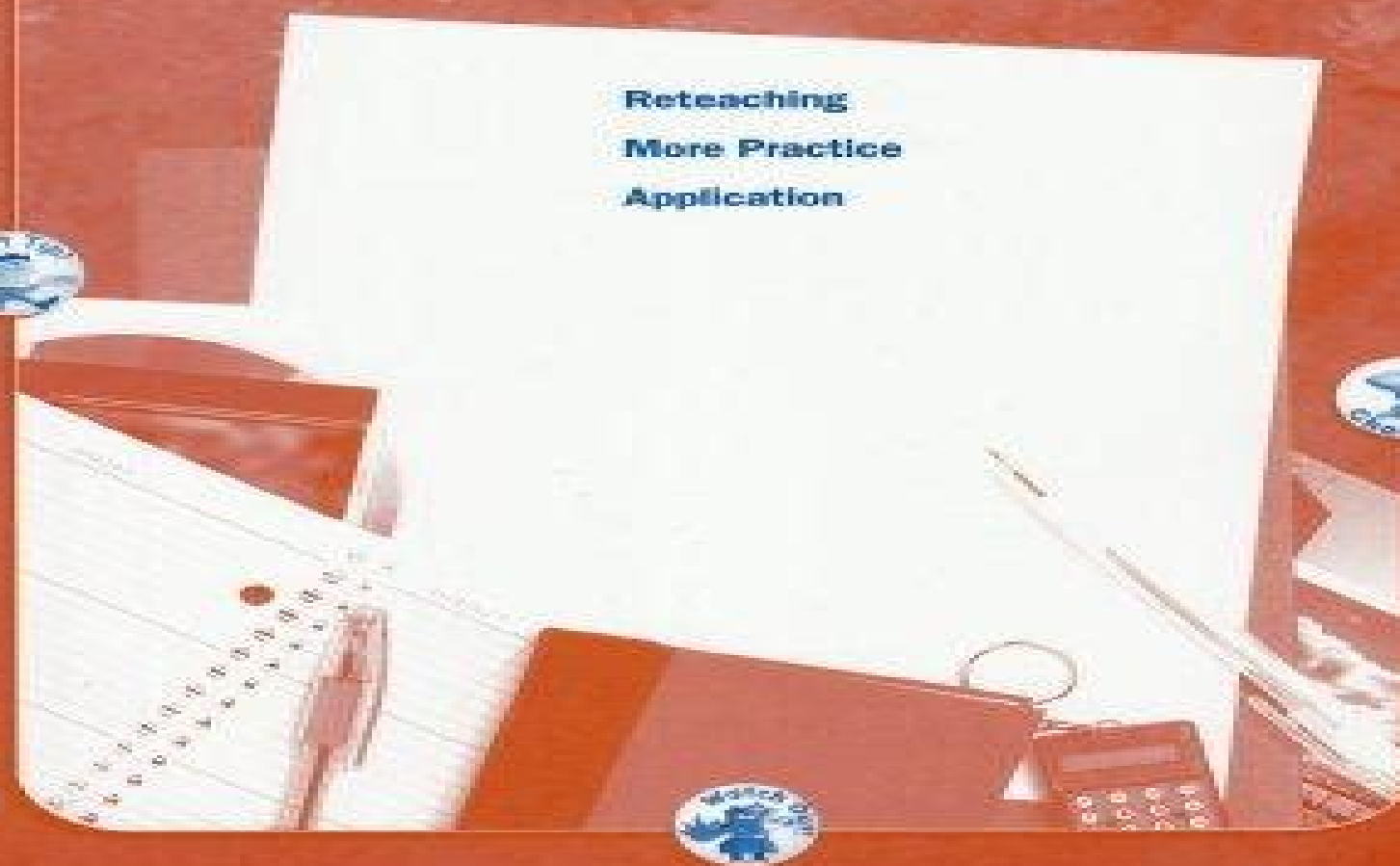


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web why dont sharks eat clowns math answers the reasons why sharks don t eat clowns are because they prefer other food we must also not forget that the clownfish location is not situated strategically

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replacement single replacement synthesis combination acid base neutralization other equation difficulty easy moderate challenging question format fill in the blank multiple choice display equations with

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web balance each of the following reactions and identify each type of reaction $\text{NaBr} + \text{Ca(OH)}_2 \rightarrow \text{CaBr}_2 + \text{NaOH}$ $\text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{NH}_4_2\text{SO}_4$ $\text{C}_5\text{H}_9\text{O}_2 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{Pb} + \text{H}_3\text{PO}_4 \rightarrow \text{H}_2\text{PbPO}_4$ $2\text{Li}_3\text{N} + \text{NH}_4\text{NO}_3 \rightarrow \text{LiNO}_3 + \text{NH}_4\text{N}$

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web mar 6 2017 1 the chemical reaction $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$ is a synthesis reaction decomposition reaction single displacement reaction double displacement reaction combustion reaction you can recognize a decomposition or analysis reaction by noticing a compound or molecule breaks into smaller chemical species 2 the chemical reaction

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web answer chemical kinetics is the branch of physical chemistry that is concerned with understanding the rates of chemical reactions it is to be contrasted with thermodynamics which deals with the direction in which a process occurs but in

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web 1 the chemical equations in model 1 contain the phase notations s l g and aq write each symbol by each meaning dissolved in water liquid solid gas aq l s g 2 based on the

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web sep 16 2022 classify a chemical reaction as a synthesis decomposition single replacement double replacement or a combustion reaction predict the products of simple reactions the chemical reactions we have described are only a tiny

sampling of the infinite number of chemical reactions possible

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web an answer key document has been included and is available for download some additional information to be aware of for teachers demonstrations 1 and 5 will not produce a chemical reaction five of the reactions will produce indicators of a chemical reaction for students to observe

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web identifying chemical reaction types key part 1 identify the type of reaction 1 $\text{Na}_3\text{PO}_4 + 3\text{KOH} \rightarrow 3\text{NaOH} + \text{K}_3\text{PO}_4$ double displacement 2 $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2\text{LiCl}$ double displacement 3 $\text{C}_6\text{H}_{12} + 9\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$ combustion 4 $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$ single displacement 5 $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ decomposition 6 $\text{P}_4 + 3\text{O}_2 \rightarrow 2\text{P}_2\text{O}_3$

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web 1 $3\text{NaBr} + \text{H}_3\text{PO}_4 \rightarrow \text{Na}_3\text{PO}_4 + 3\text{HBr}$ type of reaction double displacement 2 $3\text{Ca}(\text{OH})_2 + \text{Al}_2(\text{SO}_4)_3 \rightarrow 3\text{CaSO}_4 + 2\text{Al}(\text{OH})_3$ type of reaction double displacement 3 $3\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single displacement 4 $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$ type of reaction combustion 5 $2\text{PbSO}_4 + 2\text{PbSO}_3 + \text{O}_2$ type of reaction

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