

**Mixed Review***The Science of Physics*

Power	Prefix	Abbreviation
$10^{-14}$	atto-	a
$10^{-13}$	femto-	f
$10^{-12}$	pico-	p
$10^{-9}$	nano-	n
$10^{-6}$	micro-	$\mu$
$10^{-3}$	milli-	m
$10^{-2}$	centi-	c

Power	Prefix	Abbreviation
$10^{-1}$	deci-	d
$10^1$	deka-	da
$10^2$	hecto-	h
$10^3$	kilo-	k
$10^6$	mega-	M
$10^9$	giga-	G
$10^{12}$	tera-	T
$10^{15}$	peta-	P
$10^{18}$	exa-	E

1. Convert the following measurements to the units specified.

- a. 2.9 days to seconds \_\_\_\_\_
- b. 53 km to millimeters \_\_\_\_\_
- c. 83 cm to kilometers \_\_\_\_\_
- d. 22 mg to kilograms \_\_\_\_\_
- e. 678 kg to micrograms \_\_\_\_\_
- f.  $8.76 \times 10^7 \text{ mW}$  to gigawatts \_\_\_\_\_
- g.  $1.753 \times 10^{-13} \text{ s}$  to picoseconds \_\_\_\_\_

2. According to the rules given in Chapter 1 of your textbook, how many significant figures are there in the following measurements?

- a. 0.0045 kg \_\_\_\_\_
- b. 37.00 s \_\_\_\_\_
- c. 8 650 000 000 mi \_\_\_\_\_
- d. 0.000 000 02 17 g \_\_\_\_\_
- e. 750 in. \_\_\_\_\_
- f. 0.5000 s \_\_\_\_\_

# Holt Manual Physics Section Review Answers

**Terry C. Jones**



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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the powdered acetamide occupies 1 2 mm of the capillary tube end c

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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