Limiting Reagent

Absence of limiting reagent

$$3 H_2 + N_2 \rightarrow 2 NH_3$$
6 g 28 g 34 g

Presence of limiting reagent

1.5
$$H_2$$
 + N_2 \rightarrow 28 g

Limiting reagent

+ N₂
14 g remaining
Excess
reagent

ChemistryLearner.com

17 g

Holt Chemistry Concept Review Limiting Reactants Problems

SA Dillow

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