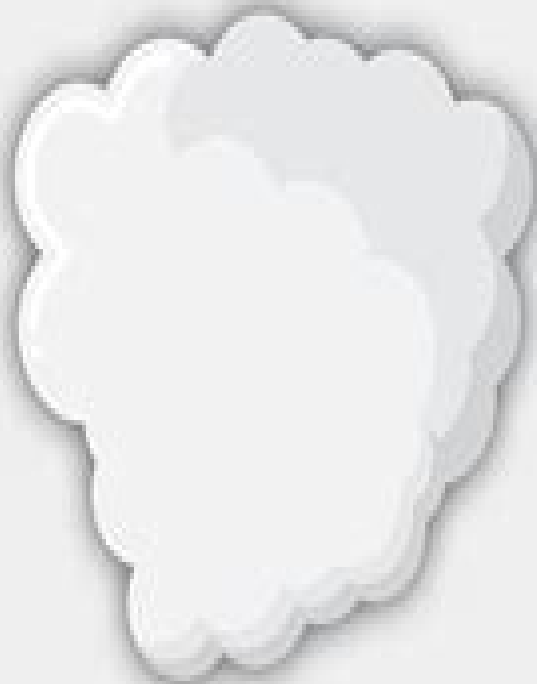


PROPERTIES OF GASES



- They have no definite shape or volume.
- Completely fill their container.
- Particles are very spread out and always in motion.
- Low density compared to liquids and solids.
- Highly compressible due to the space between particles.
- No strong attractive forces between particles.
- Change shape and volume with temperature changes.
- Expand when heated and contract when cooled.
- Easily mix with other gases.
- Their properties are described by the laws of gases.

Holt Chemistry Concept Review Characteristics Of Gases

K Morrison



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