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2 $3\text{Ca(OH)}_2 + 2\text{Al}_2\text{SO}_4 \rightarrow 4\text{Al(OH)}_3 + 3\text{CaSO}_4$ type of reaction double displacement

3 $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single displacement

4 $1\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$

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3 $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single displacement

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