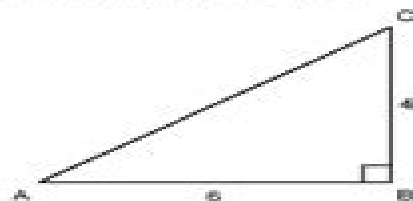


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Multiple Choice

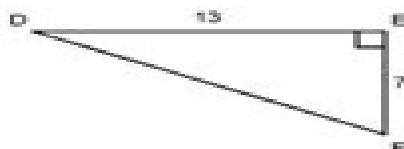
Identify the choice that best completes the statement or answers the question.

- ____ 1. Determine $\tan A$ and $\tan C$.



- a. $\tan A = 1.5$; $\tan C = 0.6$
b. $\tan A = 0.6$; $\tan C = 0.8321\dots$
c. $\tan A = 0.6$; $\tan C = 1.5$
d. $\tan A = 0.5547\dots$; $\tan C = 1.5$

- ____ 2. Determine the measure of $\angle D$ to the nearest tenth of a degree.



- a. 28.3°
b. 32.6°
c. 57.4°
d. 61.7°

- ____ 3. Determine the angle of inclination of the line to the nearest tenth of a degree.



- a. 65.0°
b. 22.9°
c. 67.1°
d. 25.0°

- ____ 4. Calculate the angle of inclination, to the nearest tenth of a degree, of a road with a grade of 19%.
a. 79.0°
b. 79.2°
c. 10.8°
d. 11.0°

- ____ 5. Determine the measure of $\angle ABD$ to the nearest tenth of a degree.

Grade Trigonometry Practice Test With Answers

Sebastian Brünink



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well as the set up for conducting the reaction students should

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction e.g. particularly nitration of methyl benzoate to give methyl m-nitrobenzoate thus determine the weight, melting point and percentage yield of the pure methyl m-nitrobenzoate abstract

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno₃ h₂so₄ in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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