

## Note:

- All questions are compulsory.
- Answer to the two sections are to be written in the same answer book.
- Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- Every new question must be started on a new page.
- Use of logarithmic table is allowed.

## CHEMISTRY: SECTION – I

Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7]

- In body centred cubic structure the space occupied is about  
(A) 68 % (B) 53 %  
(C) 38 % (D) 32 %
- For a gaseous reaction the unit of rate of reaction is  
(A)  $\text{L atm s}^{-1}$  (B)  $\text{atm mol}^{-1} \text{s}^{-1}$   
(C)  $\text{atm s}^{-1}$  (D)  $\text{mol s}$
- Which of the following compounds contains S – O as well as S – S bonds?  
(A) Sulphuric acid (B) Thiosulphuric acid  
(C) Sulphurous acid (D) Thiosulphurous acid
- Which of the following solutions shows maximum depression in freezing point?  
(A) 0.5 M  $\text{Li}_2\text{SO}_4$  (B) 1 M NaCl  
(C) 0.5 M  $\text{Al}_2(\text{SO}_4)_3$  (D) 0.5 M  $\text{BaCl}_2$
- For a chemical reaction  $\Delta S = -0.035 \text{ kJ/K}$  and  $\Delta H = -20 \text{ kJ}$ .  
(A) 5.14 K (B) 57.14 K  
(C) 571.4 K (D) 5714.0 K
- The standard e.m.f of the following cell is 0.463 V  
 $\text{Cu} | \text{Cu}_{(0.01)}^{++} || \text{Ag}_{(0.01)}^{+} | \text{Ag}$ . If  $E_{\text{Ag}}^{\circ} = 0.800 \text{ V}$ ,  
What is the standard potential of Cu electrode?  
(A) 1.137 V (B) 0.337 V  
(C) 0.463 V (D) -0.463 V
- $\text{Fe}_2\text{O}_3$  is reduced to spongy iron near the top of blast furnace by  
(A)  $\text{H}_2$  (B) CaO  
(C)  $\text{SiO}_2$  (D) CO

# Grade12013 Chemistry June Paper

**D Kirk**



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### **Table of Contents Grade12013 Chemistry June Paper**

1. Understanding the eBook Grade12013 Chemistry June Paper
  - The Rise of Digital Reading Grade12013 Chemistry June Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying Grade12013 Chemistry June Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Grade12013 Chemistry June Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from Grade12013 Chemistry June Paper
  - Personalized Recommendations

- Grade12013 Chemistry June Paper User Reviews and Ratings
- Grade12013 Chemistry June Paper and Bestseller Lists
- 5. Accessing Grade12013 Chemistry June Paper Free and Paid eBooks
  - Grade12013 Chemistry June Paper Public Domain eBooks
  - Grade12013 Chemistry June Paper eBook Subscription Services
  - Grade12013 Chemistry June Paper Budget-Friendly Options
- 6. Navigating Grade12013 Chemistry June Paper eBook Formats
  - ePub, PDF, MOBI, and More
  - Grade12013 Chemistry June Paper Compatibility with Devices
  - Grade12013 Chemistry June Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Grade12013 Chemistry June Paper
  - Highlighting and Note-Taking Grade12013 Chemistry June Paper
  - Interactive Elements Grade12013 Chemistry June Paper
- 8. Staying Engaged with Grade12013 Chemistry June Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Grade12013 Chemistry June Paper
- 9. Balancing eBooks and Physical Books Grade12013 Chemistry June Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Grade12013 Chemistry June Paper
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Grade12013 Chemistry June Paper
  - Setting Reading Goals Grade12013 Chemistry June Paper
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade12013 Chemistry June Paper
  - Fact-Checking eBook Content of Grade12013 Chemistry June Paper

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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note on time series persistence 175 7 8 4 robust asymptotic variance matrix 175 7 8 5 testing for serial correlation and

heteroskedasticity after pooled ols 176 7 8 6 feasible gls estimation under strict exogeneity 178

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