

ICP: Chemistry Review

0

out of 150

Directions: In order to earn you credit for 1st semester ICP, you must complete the following assignments. When you have completed 100% of the work below, the assignment is considered finished. When answering questions, if the answer is correct, the cell will turn color! For your reference, a periodic table is located [on the next tab](#).

Lab Equipment and Lab Safety

Directions for #1-5, identify the following pieces of lab equipment. Once all are correct, the cells will turn green. Then, decide if the statements below are true or false by choosing yes or no. Once all the answers in a column are correct, the cells will turn green.



1



2



3



4



Yes or No? (You must answer all questions in the column correctly)

Yes or No? (You must answer all questions in the column correctly)

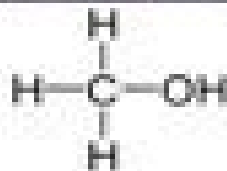
- 6. Wear safety goggles when using chemicals, hot liquids, or flames
- 7. You can try new experiments even if they are not in the directions as long as you are curious about science
- 8. Tie back long hair when using chemicals or open flames
- 9. If you spill any material, quickly clean it up with a cloth before the teacher can see that you spilled it.
- 10. When you have finished an experiment, leave your work area clean and dump all of your chemicals down the sink

- 11. It is ok to leave a burner flame alone for a few seconds while you grab some materials
- 12. In an experiment, you should have on proper lab clothing including close-toed shoes.
- 13. You should always wash your hands before an experiment, but you can decide if you need to wash your hands after an experiment
- 14. To safely smell a chemical in lab you should inhale the fumes directly from the container
- 15. Food and gum are NOT allowed in the lab area at any time.

16. Protons, Neutrons, and Electrons practice- fill in the missing information in the table below. When a row is correct, it will change color.

Atomic Symbol	Atomic Number	Protons	Neutrons	Electrons	Atomic Mass
C			6		
	11				23
		30	35		
	9		10		

Look at the diagram of methanol below to help you answer questions #17-19



- 17. How many atoms are in a compound of methanol?
- 18. What is the chemical formula of methanol?
- 19. What is the molar mass of methanol?

For #20-23, identify each as either a Physical or Chemical Change.

- 20. A phase change (ice to melting)
- 21. Mixing 2 solutions produces a substance with a new color or turns cloudy
- 22. A change that is easier to reverse
- 23. Dissolving sugar into water

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Table of Contents Icp Final Review Chemistry Id A

1. Understanding the eBook Icp Final Review Chemistry Id A
 - The Rise of Digital Reading Icp Final Review Chemistry Id A
 - Advantages of eBooks Over Traditional Books
2. Identifying Icp Final Review Chemistry Id A
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Icp Final Review Chemistry Id A
 - User-Friendly Interface
4. Exploring eBook Recommendations from Icp Final Review Chemistry Id A
 - Personalized Recommendations
 - Icp Final Review Chemistry Id A User Reviews and Ratings
 - Icp Final Review Chemistry Id A and Bestseller Lists

5. Accessing Icp Final Review Chemistry Id A Free and Paid eBooks
 - Icp Final Review Chemistry Id A Public Domain eBooks
 - Icp Final Review Chemistry Id A eBook Subscription Services
 - Icp Final Review Chemistry Id A Budget-Friendly Options
6. Navigating Icp Final Review Chemistry Id A eBook Formats
 - ePub, PDF, MOBI, and More
 - Icp Final Review Chemistry Id A Compatibility with Devices
 - Icp Final Review Chemistry Id A Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Icp Final Review Chemistry Id A
 - Highlighting and Note-Taking Icp Final Review Chemistry Id A
 - Interactive Elements Icp Final Review Chemistry Id A
8. Staying Engaged with Icp Final Review Chemistry Id A
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Icp Final Review Chemistry Id A
9. Balancing eBooks and Physical Books Icp Final Review Chemistry Id A
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Icp Final Review Chemistry Id A
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Icp Final Review Chemistry Id A
 - Setting Reading Goals Icp Final Review Chemistry Id A
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Icp Final Review Chemistry Id A
 - Fact-Checking eBook Content of Icp Final Review Chemistry Id A
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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