

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পরীক্ষার প্রয়োজক

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

$$W = 20 \text{ g}$$

$$V = 36.5 \text{ g/mol}$$

$$M = 0.5479 \text{ M}$$

$$S = 1$$

উদাহরণে উল্লিখিত X-পদার্থটিতে 10 mg এর মধ্যে ১০০ পরিমাণ ঘন ও Y-পদার্থটিতে ০.১ g ১% $\left(\frac{W}{V}\right)$ HCl নমুনা নিয়ে কাজ করা হয়েছে।

যদিও X-পদার্থটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও Y-পদার্থটি হলো মাইক্রো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় ১০০-২০০ মিলিগ্রাম সনাক্তকরণ ও পরিমাপের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (০.০১ থেকে ০.১g কঠিন বা ১-৫ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে খুব কম উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2SO_4 এর পরিমাণে কয়েকটিমাইক্রো ব্যবহার করা হয় ফলে পরিবেশে খুব কম ঘটি না।

যেহেতু কোনো কোনো ক্ষেত্রে কম কৃষিকার রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

উদাহরণ



১. উদাহরণে চলক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকটির জন্য প্রয়োজ্য হ্যাঙ্গার প্রতীকটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহরণের অংশের চলক থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার দ্রবণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪ নং প্রশ্নের উত্তর

১. বিশ্লেষণের নিমিত্ত বীমের উপর রাইডার স্থাপন করলে বীমের প্রতি সারকে অন্য যে দর পাওয়া যায়, তাকে রাইডার দ্রবক বলে।

২. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় ১০০-২০০ মিলিগ্রাম সনাক্তকরণ ও পরিমাপের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (০.০১ থেকে ০.১g কঠিন বা ১-৫ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে খুব কম উল্লেখযোগ্যভাবে কমে আসে।

৩. উদাহরণের চলকটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M দ্রব। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M দ্রব হলে পাত্র H_2SO_4 দ্রব। পাত্র H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাঙ্গার প্রতীক অঙ্কন করা হয়। পাত্র H_2SO_4 এর জন্য ক্ষয়কারী হ্যাঙ্গার প্রতীক নিয়ে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রতীক	ক্ষয়কারী (Corrosive)
আমদান	ক্ষয়কারী রাসায়নিক উপাদান কলতে যে ক্ষয়জনক রাসায়নিক দ্রব্যকে খুব কম সময় মধ্যেই ত্বক ও অন্যান্য পদার্থের ক্ষয়জনক ক্ষতি সাধিত হয়। এতে ত্বক পুড়ে যায় এবং অন্যান্য পদার্থ গলে যায়।
উদাহরণ	পাত্র কঠিন এসিডসমূহ (HCl , HNO_3 , H_2SO_4), ক্ষয়কারী ($NaOH$, KOH), H_2O_2 , অক্সিজেন লিকার প্রভৃতি দ্রব ইত্যাদি।
সতর্কতা	- পাত্র এসিডকে সোতলে রাখতে হবে। - পাত্র সংরক্ষণে, এরপর স্পর্শকৃত পদার্থকে সোতলে রাখতে হবে।

উদাহরণের চলকটি হলো 10M H_2SO_4 দ্রব। এটি এর থেকে 1M ঘনমাত্রার দ্রব প্রস্তুত করতে হবে 1ম ও ১০০ মিলিগ্রাম ব্যবহার করতে হবে।

1ম কাজঘরটি হলো 100 ml. অক্সিজেনমিত্রিক দ্রব ও এর টি হলো পিপিটি। যেহেতু অক্সিজেনমিত্রিক দ্রবের অঙ্কন 100 ml. তাই আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 দ্রব তৈরি করতে হবে।

$$V_1 S_1 = V_2 S_2$$

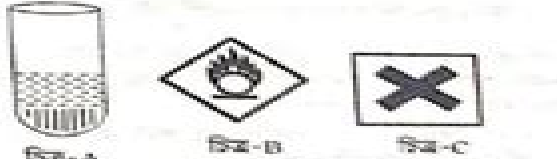
$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10mL H_2SO_4 দ্রবের সঙ্গে পানি যোগ করতে হবে (100-10) mL = 90 mL. আবার, পিপিটির সাহায্যে 10 mL দ্রবের ঘনমাত্রার দ্রব নিয়ে তার সাথে অক্সিজেনমিত্রিক দ্রব 90 mL পানি যোগ করলে আমাদের কাঙ্ক্ষিত ঘনমাত্রার দ্রব পাওয়া যাবে।

উদাহরণ



১. অক্সিজেনমিত্রিক কলকে বসে।
২. সামান্য দ্রব K₂ এর ঘন কলবে অসীম হয় বা-ব্যাখ্যা করে।
৩. A বীমের হ্যাঙ্গার থেকে উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C বীমের সাথে সম্পর্কিত রাসায়নিক দ্রব্যের ব্যবহারকালে খান্দা কৃষ্টি দ্রব কলবীর উপর সম্পর্ক বিশ্লেষণ করে। ৪ নং প্রশ্নের উত্তর

১. অক্সিজেনমিত্রিক কল এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সঞ্চেপক বা মুক্ত দ্রব্য উপাদানসমূহকে জারণ প্রক্রিয়ায় জাল থেকে ত্রুণ করে রাখার সঞ্চেপক করে।

২. সামান্য দ্রব K₂ এর ঘন কলবে অসীম বা শূন্য হতে পারে না। উল্লিখিত ব্যাখ্যা করার জন্য নিম্নের উত্তরদুই মাইক্রোপরি বিশ্লেষণ করি-
 $A + B \longrightarrow C + D$

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Rosina Ehmann



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