



NATIONAL
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2
(CHEMISTRY)
EXEMPLAR

Duration: 120
Time: 1 hour



The question paper consists of 12 pages, including 4 extra sheets.
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QUESTION 8

PHYSICAL SCIENCES PAPER 2

ACID AND BASES

GRADE 11

NOVEMBER 2020

QUESTION 8: ACID AND BASES (100 marks)

8.1. Draw a potential energy diagram for the reaction represented by

The following equation shown on the page:

a. Heat of this reaction (20)
b. Activation energy (20)

QUESTION 8 (Start on a NEW page)

8.2. Sodium hydroxide is an ionic compound with the following percentage composition:

Na 40%, O 53%, H 6%

The molar mass of the compound is 40 g mol⁻¹

8.2.1. Calculate the mass percentage of Na (20)
8.2.2. Calculate the number of moles of NaOH in 20 g (20)
8.2.3. Calculate the volume of NaOH in 20 g (20)

8.3. Concentrated sulphuric acid is a mixture of sulphuric acid and water. It is a mixture of 98% sulphuric acid and 2% water. Calculate the mass percentage of sulphuric acid in the mixture.

8.3.1. Calculate the mass percentage of sulphuric acid (20)

8.3.2. Calculate the concentration of the sulphuric acid solution (20)

8.4. Sodium hydroxide is used in the production of soap. The following reaction is used in the production of soap:

$\text{C}_{18}\text{H}_{37}\text{O}_2 + \text{NaOH} \rightarrow \text{C}_{18}\text{H}_{35}\text{O}_2 + \text{H}_2\text{O}$

Calculate the volume of hydrogen gas (H₂) that would be produced at STP if 10 g of sodium hydroxide reacts completely. (20)



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