



Motorola E2 User Manual

Yicheng Fang



Motorola E2 User Manual:

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convert to basic solution $4\text{CN}^- + 2\text{Au} + \text{CN}^- + 2\text{e}^-$ multiplied by a factor of 2 $2\text{e}^- + 2\text{H}_2\text{O} \rightarrow \text{O}_2 + 2\text{OH}^-$ 3 the final answer

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every atom of nickel oxidized but where do those electrons go

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web aug 29 2023 some points to remember when balancing redox reactions the equation is separated into two half equations one for oxidation and one for reduction the equation is balanced by adjusting coefficients and adding H^+ 2O^{2-} H^+ and e^- in this order balance the atoms in the equation apart from o and h

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6 + 2\text{H}_2\text{NO}_2 + \text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} \rightarrow 2\text{HNO}_2 + \text{C}_6\text{H}_8\text{O}_6 + 2\text{H}_2\text{O}$ choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and HNO_2 a $\text{C}_6\text{H}_8\text{O}_6$ and H_2NO_2 b HNO_2 and $\text{C}_6\text{H}_8\text{O}_6$ c $\text{C}_6\text{H}_8\text{O}_6$ and H_2O d HNO_2 and H_2O