

- Instructions: (1) Substantiate your answers with formulas for mass flow rate in cases where balance is not required.
(2) Balance is not required.
(3) Identify the type of chemical substance, its composition, single reactant, double reactant, or product.

1. A mixture of two gases with masses 100 g and 150 g is heated and expands. The mass of the mixture is:

Type: _____

Answer: 100 g + 150 g = 250 g

2. A mixture of two gases with masses 100 g and 150 g is heated.

Type: _____

Answer: 100 g + 150 g = 250 g

3. A mixture of two gases with masses 100 g and 150 g is heated. The mass of the mixture is:

Type: _____

Answer: 100 g + 150 g = 250 g

4. A mixture of two gases with masses 100 g and 150 g is heated. The mass of the mixture is:

Type: _____

Answer: 100 g + 150 g = 250 g

5. A mixture of two gases with masses 100 g and 150 g is heated. The mass of the mixture is:

Type: _____

Answer: 100 g + 150 g = 250 g

Even More Fun With Equations Answers 6 33a

Lingsheng Yao



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