

d^2sp^3 Hybridization



- d^2sp^3 hybridization enables an octahedral arrangement for six pairs of electrons surrounding a central atom.
 - Forms from two d orbitals, one s orbital, and three p orbitals.
- Examples
 - SF_6
 - XeF_4 (Xe has two lone pairs)



Evan P Silberstein 20answers Atomic Structure

Arturo Cuomo



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