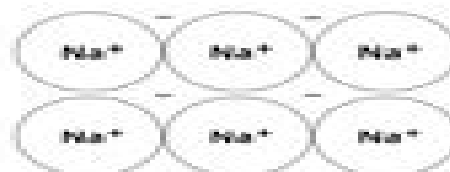


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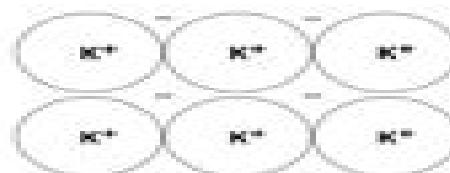
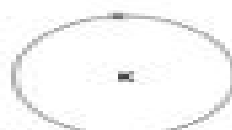
Draw diagrams to represent bonding in the each of the metals 1-5. Use six particles in each diagram.

1. Sodium, Na



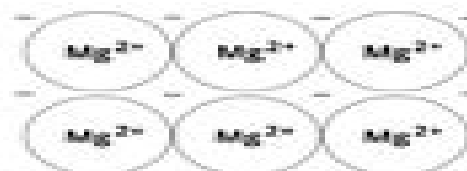
6 ions and 6 electrons

2. Potassium, K



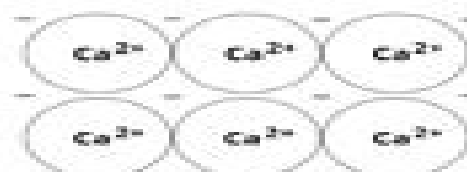
6 ions and 6 electrons

3. Magnesium, Mg



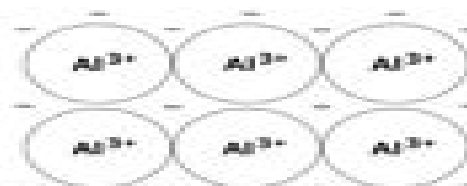
6 ions and 12 electrons

4. Calcium, Ca



6 ions and 12 electrons

5. Aluminium, Al



6 ions and 18 electrons

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Marcel A. Müller



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Crystal Structure and Chemical Bonding in Inorganic Chemistry Albrecht Rabenau, Coenraad Jozef Maria Rooymans, 1975

Chemical Misconceptions Keith Taber, 2002 Part 2 provides strategies for dealing with some of the misconceptions that students have by including ready to use classroom resources

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