

QUESTION 1

- 1.1. C ✓✓
- 1.2. D ✓✓
- 1.3. B ✓✓
- 1.4. A ✓✓
- 1.5. A ✓✓

[10]

QUESTION 2

- 2.1.
 - 2.1.1. A pure substance is a substance that cannot be separated into simpler components by physical methods. ✓✓
 - 2.1.2.1 Element ✓
 - 2.1.2.2 Consists of only one type of atom. ✓
 - 2.1.2.3 It is a combination of different metals (copper and zinc). ✓
- 2.2. Pots and pans are made of metal, because metal is a good conductor which allows heat to be transferred so that food can cook. ✓ The handles are insulators so that you do not burn your hands when you pick up a hot pot. ✓
- 2.3.1.
 - (a) Temperature ✓
 - (b) Time ✓
- 2.3.2. Boiling point is the temperature of a liquid at which its vapour pressure equals to the external (atmospheric) pressure ✓✓
- 2.3.3. The temperature remains constant because the energy is absorbed by the molecules. ✓
The strength of the intermolecular forces between the molecules decreases. ✓
The particles move faster and spaces between the particles increases. ✓
The substance changes phase from liquid to gas ✓

[15]

Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10

A Gutmann



Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10:

This book delves into Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10. Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10 is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10
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 - Chapter 3: Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10 in Everyday Life
 - Chapter 4: Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10 in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10. This chapter will explore what Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10 is, why Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10 is vital, and how to effectively learn about Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10.
 3. In chapter 2, the author will delve into the foundational concepts of Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10. The second chapter will elucidate the essential principles that need to be understood to grasp Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10 in its entirety.
 4. In chapter 3, this book will examine the practical applications of Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10 in daily life. The third chapter will showcase real-world examples of how Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10 can be effectively utilized in everyday scenarios.
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 6. In chapter 5, this book will draw a conclusion about Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Memo Of Physical Science Chemistry P2 Exemplar 2012 Grade10.

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