

Polyatomic Ions Worksheet

Polyatomic Ions

I. Name or write the formula for the following polyatomic ions:

sulfate	_____	CO_3^{2-}	_____
nitrite	_____	SO_4^{2-}	_____
perchlorate	_____	BrO_3^-	_____
hypochlorite	_____	CO_3^{2-}	_____
chlorite	_____	PO_4^{3-}	_____
phosphate	_____	ClO^-	_____
percarbonate	_____	NO_2^-	_____
chromate	_____	PO_4^{3-}	_____
hyposulfite	_____	FeO_4^{2-}	_____
permanganate	_____	SO_4^{2-}	_____
carbonate	_____		

II. Name or write the formula for the following "Type I" polyatomic ionic compounds:

barium hydrosulfide	_____	$\text{Ba}(\text{OH})_2$	_____
sodium nitrite	_____	$\text{Ca}(\text{OH})_2$	_____
ammonium chloride	_____	Ag_2SO_4	_____
calcium bisulfate	_____	MgCO_3	_____
rubidium perchlorate	_____	Sr_2FeO_4	_____
chromium sulfate	_____	$\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3$	_____
aluminum acetate	_____	SrSO_4	_____
ammonium nitrate	_____	$\text{Fe}(\text{OH})_2$	_____
magnesium hypochlorite	_____	$\text{Ca}(\text{HCO}_3)_2$	_____
silver hypochlorite	_____	Sr_2CO_3	_____
potassium cyanide	_____	$\text{Fe}(\text{OH})_3$	_____
barium iodate	_____	$\text{Ba}(\text{OH})_2$	_____

Lots Of Ionic Naming Practice Problems Key

Chao Zhang



Lots Of Ionic Naming Practice Problems Key:

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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student

demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields

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Foundations of College Chemistry Morris Hein, Susan Arena, 2013-01-01 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years Foundations of College Chemistry Alternate 14th Edition has helped readers master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis

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