

# Hydrocarbons

|                                        |                                                                                |                                                                                                                                                             |                                                                                                                                                                                                                                                                                                                                                                             |         |                                                                                                                                                                                                                                                               |
|----------------------------------------|--------------------------------------------------------------------------------|-------------------------------------------------------------------------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|---------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Methane                                | General formula for an alkane                                                  | General formula for an alkene                                                                                                                               | Ethene                                                                                                                                                                                                                                                                                                                                                                      | Propene | Butene                                                                                                                                                                                                                                                        |
| Ethane                                 | How to test between an alkene and alkane                                       | Alkane or alkene?<br><br>C <sub>2</sub> H <sub>6</sub><br>C <sub>3</sub> H <sub>8</sub><br>C <sub>4</sub> H <sub>10</sub><br>C <sub>5</sub> H <sub>12</sub> | Show how ethene is polymerised                                                                                                                                                                                                                                                                                                                                              |         |                                                                                                                                                                                                                                                               |
| Propane                                | Definition of a hydrocarbon                                                    |                                                                                                                                                             | What is meant by saturated/unsaturated                                                                                                                                                                                                                                                                                                                                      |         |                                                                                                                                                                                                                                                               |
| Butane                                 | How ethanol is produced...<br>...from ethane<br><br><br><br>...by fermentation |                                                                                                                                                             | Draw the monomer that makes this polymer<br>$  \begin{array}{cccccc}  \text{Br} & \text{H} & \text{Br} & \text{H} & \text{Br} & \text{H} \\    &   &   &   &   &   \\  \text{H}-\text{C}- & \text{C}- & \text{C}- & \text{C}- & \text{C}- & \text{C}-\text{H} \\    &   &   &   &   &   \\  \text{H} & \text{H} & \text{H} & \text{H} & \text{H} & \text{H}  \end{array}  $ |         |                                                                                                                                                                                                                                                               |
| A way to remember the order of alkanes |                                                                                |                                                                                                                                                             | Describe the process of cracking<br>Process:<br><br><br>Types of products produced:<br><br>Conditions:<br><br>Why it is done:                                                                                                                                                                                                                                               |         | Complete the equations for cracking<br>$\text{C}_{10}\text{H}_{22} \rightarrow \text{C}_2\text{H}_6 + \text{_____} + \text{C}_6\text{H}_{14}$<br>$\text{C}_8\text{H}_{18} \rightarrow \text{C}_3\text{H}_8 + \text{_____}$<br>The types of products produced: |

# Hydrocarbon Compounds Core Teaching Resources

**Jay K. Hackett**



## **Hydrocarbon Compounds Core Teaching Resources:**

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effects of each hazardous material Comprehensive coverage of individual toxic elements including Asbestos Alar Lead Mercury Coverage of hazardous material groups such as Pesticides Food additives Nitrogen compounds More general articles such as Evaluation and testing of carcinogens Transport of pollutants Inhalation toxicology

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In a global driven by information and connectivity, the power of words has be much more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Hydrocarbon Compounds Core Teaching Resources**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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