Solutions: Metathesis Reactions and Net Ionic Equations

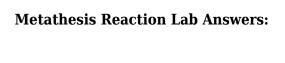
9 Pre-lab Questions

Before beginning this experiment in the laboratory, you should be able to answer the following questions.

- Write molecular, complete ionic, and net ionic equations for the reactions that occur, if any, when solutions
 of the following substances are mixed:
 - (a) nitric acid and potassium carbonate
 - (b) sodium bromide and lead nitrate
 - (c) acetic acid and calcium hydroxide
 - (d) calcium nitrate and sodium sulfate
 - (e) ammonium chloride and lithium hydroxide
- Which of the following are not water-soluble: Ba(NO₃)₂, FeCl₃, CuCO₃, CuSO₄, ZnS, and ZnSO₄?
- 3. Write equations for the decomposition of H2CO3 and H2SO3.
- 4. At what temperature (from your graph) do KNO3 and NaCl have the same molar solubility?

Metathesis Reaction Lab Answers

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