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ELECTRIC & GAS DRYER SERVICE MANUAL

CAUTION

READ THIS MANUAL CAREFULLY TO DIAGNOSE
PROBLEMS PROPERLY AND TO PROVIDE
QUALITY SERVICE SAFELY ON THESE DRYERS.



MODEL : DLEX3900* DLGX3901*
DLEX3700* DLGX3701*
DLE3500* DLG3501*

Lg Dryer User Manual

Richard Bailey



Lg Dryer User Manual:

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web the coefficient in a balanced equation is an idea the concept of telling the chemist that if the atoms molecules and compounds are balanced there are balanced amounts of the atoms molecules and compounds on the other side in the product

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the chemical equation after this you can list down the products on the right hand side of the chemical equation

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 $1\text{ N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
 $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $4\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + 5\text{PbCl}_2$
 $6\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 + 6\text{KBr}$
 $1\text{Al} + 2\text{SO}_4 + 3\text{H}_2 + 7\text{CH}_4 + 2\text{O}_2 \rightarrow 1\text{CO}_2 + 2\text{H}_2\text{O} + 8\text{H}_2\text{O} + 1\text{C}_3\text{H}_8 + 5\text{O}_2 + 3\text{CO}_2 + 4\text{H}_2\text{O}$

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 $1\text{O}_2 + 3\text{H}_2 \rightarrow 3\text{H}_2\text{O}$
 $2\text{H}_2\text{O} + 2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + 2\text{O}_2 + 3\text{H}_2$
 $2\text{O}_2 + 2\text{H}_2\text{O} + 2\text{O}_2 + 4\text{Rb} + 2\text{H}_2\text{O} + 2\text{RbOH} + 2\text{H}_2 + 5\text{Cl}_2 + 2\text{KBr} + \text{KCl} + \text{Br}_2 + 6\text{N}_2 + 2\text{H}_2 + 2\text{NH}_3 + 7\text{N}_2\text{F}_4 + 3\text{Ca} + \text{N}_2\text{O} + \text{C}_2\text{H}_5\text{OH}$

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 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + \text{PbCl}_2$
 $2\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$
 $3\text{K}_2\text{SO}_4 + 6\text{KBr} + 1\text{Al}_2\text{SO}_4 + 3\text{CH}_4 + 2\text{O}_2 \rightarrow 1\text{CO}_2 + 2\text{H}_2\text{O}$

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 $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
 $2\text{Fe} + 2\text{O}_3 \rightarrow 2\text{FeO} + 3\text{H}_2\text{SO}_4 + 1\text{Fe}_2\text{SO}_4 + 3\text{H}_2\text{O} + 1\text{C}_4\text{H}_{10} + 3\text{H}_2\text{O} + 2\text{C}_2\text{H}_4\text{O}_2$

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 $38\text{Fe} + 2\text{O}_3 + 3\text{CO} + 2\text{Fe}_3\text{CO}_2 + 39\text{H}_2 + 3\text{PO}_4 + 3\text{BaO} + 3\text{Ba} + 3\text{PO}_4 + 2\text{H}_2 + 40\text{Mg} + 4\text{MgNH}_4 + 4\text{PO}_4 + 2\text{P}_2 + 7\text{N}_2 + 3\text{H}_2 + 41\text{Al} + 3\text{H}_2 + 4\text{Al}_2\text{SO}_4 + 3\text{H}_2 + 42\text{P}_4 + 10\text{HCl} + \text{POCl}_3 + 3\text{HPO}_3 + 43\text{HNO}_2 + 2\text{H}_2 + 3\text{O}_2 + 4\text{HNO}_2$

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