

Organic Chemistry

01. What is the IUPAC name of $\text{HC} \equiv \text{C} - \text{CH}_2 - \text{CH} = \text{CH}_2$?
- A. Pent 1 ene 4 yne
B. Pent 1 yne 4 ene
C. 1 pentyne 4 yne
D. 1 pent 1 yne
02. Which one of the following have chiral carbon?
- A. $\text{CH}_3\text{CH}_2\text{COOH}$
B. $\text{CH}_3\text{CH}(\text{NH}_2)\text{COOH}$
C. $\text{CH}_2(\text{NH}_2)\text{COOH}$
D. $\text{C}_2\text{H}_5\text{OH}$
03. Which of the following shows geometric isomerism?
- A. $\text{ClCH} = \text{CHCl}$
B. $\text{Cl}_2\text{C} = \text{CH}_2$
C. $\text{CH}_2 = \text{CH}_2$
D. $\text{CH}_2 = \text{CHCl}$
04. Which one is a homologous series?
- A. Ethane, methane, propane
B. Ethane, propane, butane
C. 1-hexene, 2-hexene, 3-hexene
D. Ethane, propane, butane
05. Presence of which group shows Bayer's test despite being unsaturated compound?
- A. $-\text{CN}$
B. $-\text{SO}_3\text{H}$
C. $-\text{CHO}$
D. $-\text{COOH}$
06. Which one is the main ingredient of Dettol?
- A. Chlorobenzene
B. Chloroxylenol
C. Benzenehexachloride
D. Chloral
07. What does ethyl iodide produce reacting with KOH (alcoholic solution)?
- A. Ethyne
B. Ethene
C. Ethanol
D. Ethane
08. Which one reacts immediately with Lucas' reagent?
- A. Butan 1 ol
B. 2 methylpropan 2 ol
C. 2 methylpropan 1 ol
D. Butan 2 ol
09. How many sigma and pi bond are there in Propyne?
- A. 6 sigma, 2 pi
B. 3 sigma, 2 pi
C. 2 sigma, 2 pi
D. 2 sigma, 4 pi
10. Which group is meta indicator in aromatic replacement reaction?
- A. $-\text{NO}_2$
B. $-\text{OH}$
C. $-\text{Cl}$
D. $-\text{CH}_3$
11. In which one Clemensen reduction didn't occur?
- A. Ethane from ethanal
B. Propane from propanone
C. Toluene from benzaldehyde
D. Ethanol from ethanoic acid

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the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner

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