

# Analytical Chemistry MCQs: TYBSC SEM III

## 1] Gravimetric Analysis

1. If the solubility of calcium fluoride in water is  $S$  moles per litre, its solubility is given by

- (a)  $2S$  (b)  $2S^2$   
(c)  $4S^3$  (d)  $4S^2$

2. Correct representation of solubility product of  $\text{Ca}_3(\text{PO}_4)_2$  is

- (a)  $[\text{Ca}^{2+}][\text{PO}_4^{3-}]^2$  (b)  $[\text{Ca}^{2+}][\text{PO}_4^{3-}]$   
(c)  $[\text{Ca}][\text{PO}_4]$  (d) none of these

3. Dimethyl glyoxime is a reagent specific for the metal ion.

- (a) Al (b) Ni  
(c) Fe (d) Cu

4. The solubility product expression for  $\text{Hg}_2\text{Cl}_2$  is

- (a)  $[\text{Hg}^{2+}][\text{Cl}]$  (b)  $[\text{Hg}_2^{2+}][\text{Cl}]^2$   
(c)  $[\text{Hg}_2^{2+}][2\text{Cl}]$  (d)  $[\text{Hg}^{+1}][\text{Cl}]$

5. A solution is called saturated if

- (a) ionic product < solubility product.  
(b) ionic product > solubility product.  
(c) ionic product = solubility product.  
(d) none of these

6. Solubility of alcohol in water is due to

- (a) ionic bonding  
(b) covalent bonding  
(c) Metallic bonding  
(d) Hydrogen bonding

7. Precipitation takes place when the product of concentration of ions

- (a) Exceeds their solubility product  
(b) Is less than their solubility product  
(c) Equal to their solubility product  
(d) None of these

# Mcq Of Analytical Chemistry

**Rosina Ehmman**



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