

Name: _____ Date: _____ Period: _____

Midterm Exam Study Guide

d. Isotopes

- i. Isotopes differ in their _____ and number of _____

e. Average atomic mass Formula:

f. Ions

- i. _____ have lost electrons and have a _____ charge
ii. _____ have gained electrons and have a _____ charge

g. Group Numbers be able to locate them

- i. Alkali Metals
ii. Alkaline earth Metals
iii. Transition Metals
iv. Inner Transition Metals
v. Halogens
vi. Noble Gases

h. Periodic Trends Give your notes for reference (the charts for each of these)

- i. Atomic Radius/Size
ii. Ionic Radius/Size
iii. Ionization Energy
iv. Electronegativity

i. Valence Electrons be able to identify # of valence electrons and charge for each group

- i. Lewis-Dot structures

2. Nuclear Decay and Half-life

- a. 0_0e or ${}^0_{-1}e$ is called an _____ particle which has the weakest radiation.
b. 4_2He or 4_2He is called an _____ particle which has moderate radiation and is an alpha.
c. ${}^0_{-1}e$ is called an _____ particle which has the strongest radiation and highest energy.
d. Half-life
i. Definition: _____
ii. Know how to read a graph and calculate half-life

3. Electron Configurations

- a. Be able to draw them/identify them from another student
b. Hund's Rule
c. Pauli Exclusion Principle
d. Aufbau Principle

4. Ionic Compounds

- a. Ionic compounds are made of _____ + _____
b. Their overall charge is _____
c. Know how to use Lewis-Dot Structures to draw bonds between cation and anion

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Resources in Education ,1985 **Subject Guide to Books in Print** ,1971 **Dissertation Abstracts**

International ,2004 *Chemistry - the Physical Setting* ,2004 **AP Chemistry Exam Success** Lewis Morris,2018-07-14

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Study Guide for Chemistry H LeMay, Theodore Brown, Bruce Bursten, Catherine Murphy, Patrick Woodward, Matthew Stoltzfus, 2017-03-21 This guide assists students through the text material with chapter overviews learning objectives a review of key terms as well as self tests with answers and explanations This student guide also features MCAT practice questions

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AP Chemistry Test Secrets, 2010

Student Study Guide/Solutions Manual to accompany General, Organic & Biological Chemistry Erin Smith Berk, Janice Smith, 2012-02-10 Each chapter of the Student Study Guide begins with a chapter review tied to the chapter goals in the text Next Sample problems are supplied and stepped out through the solution for each type of problem covered in the chapter A Self Test serves up fill in the blank exercises to assess learning with answers supplied at the end of the chapter Finally chapters end with the solutions for all of the in chapter problems as well as for the odd numbered end of chapter problems

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