

PART - I

15 X 1 = 15

- I Answer all the questions.**
- Which of the following is / are true with respect to Carbon - 12
a) relative atomic mass is 12u b) oxidation number of Carbon is +4 in all its compounds
c) mole of Carbon - 12 contain 6.022×10^{22} carbon atoms d) all of these
 - How many electrons in an atom with atomic number 105 can have $(n + l) = 8$
a) 30 b) 17 c) 15 d) unpredictable
 - Among the following which is isoelectronic with O^{2-} ion,
a) Cl^- b) F^- c) Ca^{2+} d) N^{2-}
 - The cause of permanent hardness of water is due to
a) $Ca(HCO_3)_2$ b) $Mg(HCO_3)_2$ c) $CaCl_2$ d) $MgCO_3$
 - Sodium is stored in
a) Alcohol b) Water c) Kerosene d) None of these.
 - It temperature and volume of an ideal gas is increased to twice its values, the initial pressure P becomes
a) 4p b) 2p c) p d) 3p
 - Heat of combustion is always
a) Positive b) negative c) zero d) either positive or negative
 - In the reaction, $Fe(OH)_3(s) \rightleftharpoons Fe^{3+}(aq) + 3OH^-(aq)$ if the concentration of OH^- ions is decreased by $1/4$ times, then the equilibrium concentration of Fe^{3+} will
a) not changed b) also decreased by $1/4$ times
c) increase by 4 times d) increase by 64 times
 - According to Raoult's law, the relative lowering of vapour pressure for a solution is equal to
a) mole fraction of solvent b) mole fraction of solute
c) number of moles of solute d) number of moles of solvent
 - Number of sigma (σ) and pi (π) bonds in acetylene is
a) 2σ & 2π b) 3σ & 3π c) 2σ & 3π d) 3σ & 2π
 - Sodium nitroprusside reacts with sulphide ion to give a purple colour due to the formation of
a) $[Fe(CN)_5 NO]^{3-}$ b) $[Fe(NO)_5 CN]^+$ c) $[Fe(CN)_5 NOS]^{4-}$ d) $[Fe(CN)_5 NOS]^{3-}$
 - What is the hybridisation state of benzyl carbonium ion?
a) sp^2 b) sp^2d c) sp^3 d) sp^2d
 - is used for welding and cutting metals.
a) Ethylene b) Propylene c) Poly ethylene d) Oxy - acetylene
 - What should be the correct IUPAC name of diethyl chloromethane?
a) 3 - chloro pentane b) 1 - chloropentane
c) 1 - chloro - 1, 1, diethyl methane d) 1 - chloro - 1 - ethylepropane
 - The pH of normal rain water is
a) 6.5 b) 7.5 c) 5.6 d) 4.6

PART - II

6 x 2 = 12

- II Answer any six. (q.No. 24 is compulsory).**
- Define modern periodic law.
 - What is water - gas shift reaction?

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