Directions: Match each term in Column I with its description in Column II. Write the letter of the correct term in the space provided.

# Column I binary compound chemically stable 3. nonpolar molecule \_\_\_\_\_ 4. ion \_\_\_\_ 5. ionic bond 6. polar molecule 7. oxidation number 8. chemical formula 9. covalent bond \_\_\_\_\_ 10. hydrate ----- 11. chemical bond \_\_\_\_\_ 12. polyatomic ion

# Column II

a. number that indicates how many electrons an atom must gain, lose, or share to become stable

Class

- b. shorthand that tells what elements a compound contains and the exact number of atoms of each element in a unit of the compound
- c. positively or negatively charged, covalently bonded group of atoms
- compound composed of two elements
- e. describes an atom that has a full outermost energy level
- f. molecule that has a slightly positive end and a slightly negative end
- g. the attraction that forms between atoms when they share electrons
- h. the force that holds atoms together in a compound
- a compound that has water chemically attached to it
- j. the force of attraction between the opposite charges of the ions in an ionic compound
- k. molecule made of two identical atoms that share the electrons equally
- a charged particle that has either more or fewer electrons than protons

# **Gpscience Physical Science Chemical Bonds Answer**

N. N. Sirota

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Linus Pauling - Selected Scientific Papers (In 2 Volumes) - Volume 1 Barclay Kamb, Linda Pauling Kamb, Carl Alexander Kamb, Peter Jeffress Pauling, Linus Pauling Jr, 2001-11-02 Linus Pauling wrote a stellar series of over 800 scientific papers spanning an amazing range of fields some of which he himself initiated This book is a selection of the most important of his writings in the fields of quantum mechanics chemical bonding covalent ionic metallic and hydrogen bonding molecular rotation and entropy protein structure hemoglobin molecular disease molecular evolution the antibody mechanism the molecular basis of anesthesia orthomolecular medicine radiation chemistry biology and nuclear structure Through these papers the reader gets a fresh unfiltered view of the genius of Pauling's many contributions to chemistry chemical physics molecular biology and molecular medicine Chemical Bonds in Solids Academician N. N. Sirota, 2012-12-06 The present four volumes published under the collective title of Chemical Bonds in Solids are the translation of the two Russian books Chemical Bonds in Crystals and Chemical Bonds in Semiconductors These contain the papers presented at the Conference on Chemical Bonds held in Minsk between May 28 and June 3 1967 together with a few other papers denoted by an asterisk which have been specially incorporated Earlier

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