

B.E. (Mechanical Engineering / Power Engineering) Fourth Semester (C.B.S.)
Engineering Thermodynamics

P. Pages : 4

Time : Three Hours



TKN/KS/16/7372/7396

Max. Marks : 80

- Notes :
1. All questions carry marks as indicated.
 2. Solve Question 1 OR Questions No. 2.
 3. Solve Question 3 OR Questions No. 4.
 4. Solve Question 5 OR Questions No. 6.
 5. Solve Question 7 OR Questions No. 8.
 6. Solve Question 9 OR Questions No. 10.
 7. Solve Question 11 OR Questions No. 12.
 8. Assume suitable data whenever necessary.
 9. Illustrate your answers whenever necessary with the help of neat sketches.
 10. Use of non programmable calculator is permitted.
 11. Use of steam table & Mollier chart is permitted.
 12. The solutions must be supported with appropriate P-v, T-s, h-s diagrams.

- a) Explain what you understand by thermodynamics equilibrium. 3

b) A platinum wire is used as a resistance thermometer. The wire resistance was found to be 10 ohm and 16 ohm at ice point and steam point respectively, and 30 ohm at sulphur boiling point of 444.6°C. Find the resistance of the wire at 500°C, if the resistance varies with the temperature by the relation $R = R_0(1 + \alpha t + \beta t^2)$. 6

c) Show that work is a path functions, and not a property. 4

OR
- a) Explain the terms : 6

 - i) specific heat at constant volume and
 - ii) specific heat at constant pressure.

Also prove that $C_p - C_v = R$.

b) A gas in a piston-cylinder assembly undergoes an expansion process for which the relation between pressure and volume is given $P^{1/n} = \text{constant}$. The initial pressure is 0.5 MPa, the initial volume is 0.15 m^3 and the final volume is 0.25 m^3 . Determine the work transfer for the process in kJ if 7

 - i) $n = 0$
 - ii) $n = 1.0$ and
 - iii) $n = 1.5$

Draw the process on P-V diagram.
- a) What was the contribution of J. P. Joule in establishing the first law of thermodynamics? What is a PMM1 and why is it impossible. 4

b) Write the first law of thermodynamics for open and closed system. Explain terms involved in it. 3

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