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APJ ABDUL KALAM TECHNOLOGICAL UNIVERSITY

First Semester B.Tech Degree Regular and Supplementary Examination December 2020 (2019 scheme)

Course Code: CYT100**Course Name: ENGINEERING CHEMISTRY
(2019-Scheme)**

Max. Marks: 100

Duration: 3 Hours

PART A**Answer all questions, each carries 3 marks.**

Marks

- | | | |
|----|--|-----|
| 1 | What will be the standard electrode potential of $\text{Ni}^{2+} / \text{Ni}$ electrode if the cell potential of the cell $\text{Ni} / \text{Ni}^{2+}(1\text{M}) // \text{Cu}^{2+}(0.1\text{M}) / \text{Cu}$ is 0.59 V at 25 °C?
$E^\circ_{\text{Cu}^{2+}/\text{Cu}} = 0.34 \text{ V}$ | (3) |
| 2 | Briefly explain the principle of electroless plating. | (3) |
| 3 | Give the mechanism of interaction of electromagnetic radiation with oscillating dipole. | (3) |
| 4 | State Beer-Lambert's law and write the differential form. | (3) |
| 5 | Write any three differences of TGA and DTA. | (3) |
| 6 | Explain sol-gel method for the synthesis of nano particles | (3) |
| 7 | Determine the configuration of the following alkenes as E or Z: | (3) |
| | <div style="display: flex; justify-content: space-around; align-items: center;"> <div style="text-align: center;"> $\begin{array}{c} \text{H}_3\text{C} \quad \quad \text{CH}_2\text{OH} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ \text{H} \quad \quad \text{Cl} \end{array}$ <p>A</p> </div> <div style="text-align: center;"> $\begin{array}{c} \text{C}_6\text{H}_5 \quad \quad \text{COOH} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ \text{H} \quad \quad \text{CH}_3 \end{array}$ <p>B</p> </div> </div> | |
| 8 | Mention any three advantages of OLEDs over LED and LCD. | (3) |
| 9 | Explain disinfection by chlorination. | (3) |
| 10 | Compare BOD and COD. | (3) |

PART B**Answer one full question from each module, each question carries 14 marks****Module-I**

- | | | |
|----|--|------|
| 11 | a) Explain the mechanism of electrochemical corrosion in different environmental conditions. | (10) |
| | b) How is the cell constant of a conductivity cell measured? | (4) |

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Table of Contents Engineering Chemical Nquestion Paper And Memo

1. Understanding the eBook Engineering Chemical Nquestion Paper And Memo
 - The Rise of Digital Reading Engineering Chemical Nquestion Paper And Memo
 - Advantages of eBooks Over Traditional Books
2. Identifying Engineering Chemical Nquestion Paper And Memo
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Engineering Chemical Nquestion Paper And Memo
 - User-Friendly Interface
4. Exploring eBook Recommendations from Engineering Chemical Nquestion Paper And Memo
 - Personalized Recommendations
 - Engineering Chemical Nquestion Paper And Memo User Reviews and Ratings
 - Engineering Chemical Nquestion Paper And Memo and Bestseller Lists
5. Accessing Engineering Chemical Nquestion Paper And Memo Free and Paid eBooks

- Engineering Chemical Nquestion Paper And Memo Public Domain eBooks
- Engineering Chemical Nquestion Paper And Memo eBook Subscription Services
- Engineering Chemical Nquestion Paper And Memo Budget-Friendly Options
- 6. Navigating Engineering Chemical Nquestion Paper And Memo eBook Formats
 - ePub, PDF, MOBI, and More
 - Engineering Chemical Nquestion Paper And Memo Compatibility with Devices
 - Engineering Chemical Nquestion Paper And Memo Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Engineering Chemical Nquestion Paper And Memo
 - Highlighting and Note-Taking Engineering Chemical Nquestion Paper And Memo
 - Interactive Elements Engineering Chemical Nquestion Paper And Memo
- 8. Staying Engaged with Engineering Chemical Nquestion Paper And Memo
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Engineering Chemical Nquestion Paper And Memo
- 9. Balancing eBooks and Physical Books Engineering Chemical Nquestion Paper And Memo
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Engineering Chemical Nquestion Paper And Memo
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Engineering Chemical Nquestion Paper And Memo
 - Setting Reading Goals Engineering Chemical Nquestion Paper And Memo
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Engineering Chemical Nquestion Paper And Memo
 - Fact-Checking eBook Content of Engineering Chemical Nquestion Paper And Memo
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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