

1. There are four types of orbitals:
- s :** shaped like a sphere  
An E level can contain only 1 s orbital, making up the "s sublevel".
  - p :** shaped like dumbbells  
An E level can contain 3 p orbitals, making up the "p sublevel".
  - d :** shaped like double dumbbells  
An E level can contain 5 d orbitals, making up the "d sublevel".
  - f :** too complex to draw or describe  
An E level can contain 7 f orbitals, making up the "f sublevel".
2. Each orbital can hold a maximum of 2 electrons. Since both electrons have a negative charge, they repel. What keeps them from flying apart?  
*Each electron spins on its axis. One spins clockwise and the other spins counter-clockwise. When charged particles spin, they act like tiny magnets. Since the two electrons spin in opposite directions, one acts like the north pole of a magnet and the other acts like the south pole. This makes the electrons attract.*
3. Since each orbital can hold 2 electrons:
- The "s sublevel" can hold 2 electrons.
  - The "p" sublevel can hold 6 electrons.
  - The "d" sublevel can hold 10 electrons.
  - The "f" sublevel can hold 14 electrons.

We use this notation to describe an electron:



How are electrons distributed within a sublevel?

*According to Hund's Rule, each orbital within a sublevel is half-filled before any is filled.*

# Electron Distributions Review Answer Key

**S. Rao Vallabhaneni**



## **Electron Distributions Review Answer Key:**

Chemistry: Core Concepts, 3rd Edition Allan Blackman, Daniel Southam, Gwendolyn Lawrie, Natalie

Williamson, Christopher Thompson, 2024-01-09 The third edition of Chemistry Core Concepts Blackman et al has been developed by a group of leading chemistry educators for students entering university with little or no background in chemistry Available as a full colour printed textbook with an interactive eBook code this title enables every student to master concepts and succeed in assessment Lecturers are supported with an extensive and easy to use teaching and learning package *Wiley CIA Exam Review 2023, Part 2* S. Rao Vallabhaneni, 2022-11-15 Banish test day anxiety with the ultimate

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## **Electron Distributions Review Answer Key** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Electron Distributions Review Answer Key**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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