

CHM 101

Electron Structure Worksheet

Objectives:

- Explain the nature of light and how it reveals the electron structure of the atom.
- Write the electron structure of elements
- Relate electron structure to the periodic table

1. Light is a form of what type of energy?
 - Give 2 examples of other forms of this type of energy.
2. What is the meaning of wavelength and frequency of light?
3. White light produces a continuous spectrum. Elements that are excited produce a line spectrum. What is a line spectrum?
4. Different metals emit different colored light in flame tests. This light originates from inside the metal atoms. Explain what was happening inside these atoms that causes the various colors of light. Be specific
5. What is an orbital?
6. List the orbitals that exist in the following energy levels. Indicate the number of these orbitals.
 $n = 1$, $n = 2$, $n = 3$, $n = 4$
7. Indicate the electron configuration of these elements using the shorthand notation and the orbital diagrams.



For example - helium: $1s^2$, $1s$

Elements: Li, C, Mg, Fe, Kr

8. In this periodic chart, indicate the outer orbitals being filled with electrons in the rectangles.

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9. Why do elements in the same group have similar properties?

Electron Configuration Worksheet Packet

IM Harris



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Progressive Science Class IX Chandan Sukumar Sengupta, This hand book is meant for students having a plan for preparing Pre Medical Board Examinations and also a plan for optng competitive examinations like NEET BDS and other such entrance examinations There will be sa series of such publications which are advanced for covering different content areas of the study These are merely a reparatory study meant primarily for equipping an individual for the forthcoming challenges Contents are designed on the basis of the recommendations made by the Curriculum Framework Proposal of NCERT for Students aspiring for National Entrance Test meant for seeking admission in Under Graduate Medical Institutions There are twn such volume for clearing the fundamental concepts of Science related doubts This book has been published with all reasonable efforts taken to make the material error free after the consent of the author No part of this book shall be used reproduced in any manner whatsoever without written permission from the author except in the case of brief quotations embodied in critical articles and reviews This workbook is meant for students having eagerness for improving in later course of study in the field of science and technology It will also expose an individual to some higher challenges of studies

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