	rite the abbreviated electron configurations of the following? ii. B
iii.O	
v. Cl	
vii. Se ³	viii. Cr ²⁺
ix. I	x. Ag
2. Identify the element v	with the following electron configuration
i. 1s² 2s² 2p° 3s² 3p° 4s² 3c	130 4p4
ii. 1s² 2s² 2p° 3s² 3p° 4s² 3	4p° 5s²
iii. 1s² 2s² 2p° 3s² 3p° 4s² :	3d30 4p6 5s1 4d7
iv. 1s2 2s2 2p6 3s2 3p6 4s2 3	3d10 4p6 5s2 4d10 5p6 6s2 4f14 5d10
v. 1s2 2s2 2p4 3s2 3p4 4s2 3	3d10 4p° 5s2 4d10 5p° 6s2 4f14 5d10 6p° 7s1
3. Determine if the follows: 1s ² 2s ² 2p ⁶ 3s ² 3p ⁶ 4s ² 4dii. 1s ² 2s ² 2p ⁶ 3s ³ iii. [Rn] 7s ² 5f ⁶ 6d ² iv. [Ar] 5s ² 4dil ⁶ 5p ³ v. [Xe] 6s ² 4fil ⁶	wing electron configurations are correct: d¹0 4pº 5s¹
4. Identify the following	elements
i. 1s ² 2s ² 2p ²	
ii. 1s ² 2s ² 2p ⁶	
iii. [Ar] 4s2 3d10 4p3	
iv. [Kr] 5s2 4d1	
v. [Ne] 3s2 3p3	

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