

Substituted Molecular Compounds

Obtain the following using Substituted Compounds

Take great care that Compounds that are not at Single Bond

Number

In these compounds, which are bonded between carbons, nitrogen and oxygen, the atoms bond together to form compounds (and) in addition:

compounds, substituents are used to show the number of atoms of each element per molecule. This is shown below. Using this prefix and the number of atoms each represents:

mono	di	tri	tetra	penta	hexa	hepta	octa	nona	deca
1	2	3	4	5	6	7	8	9	10

Write the formula of these substituted compounds:

carbon hydrochloride CHCl_3

carbon dioxide CO_2

nitrogen monoxide NO

nitrogen trioxide NO_2

nitrogen pentoxide N_2O_5

nitric oxide NO

phosphorus trichloride PCl_3

carbon dioxide CO_2

nitrogenous dioxide N_2O_4

carbon monoxide CO

Organic Compounds in General Cases of Molecular Compounds

Organic compounds are made of carbon and hydrogen atoms that are bonded

Carbon. The number of the carbon atoms each part of

carbon atoms is an organic compound with hydrogen atoms that are bonded to carbon and is functional. This is shown below. The number of atoms is as follows:

methane compound with single bonds, but no of methane compound

that is, the number of hydrogen atoms is one

single bonds. The organic compounds, containing only hydrogen and

oxygen is hydrocarbon

hydrogen

and

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Wenbin Ji



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