

# 9 • Bonding & Molecular Structure

## PRACTICE TEST

1. The correct Lewis symbol for ground state carbon



2. The correct Lewis symbol for ground state aluminum is



3. Using the picture below, what process corresponds to the lattice energy?



4. Which of the following favors formation of an ionic compound?

- (a) low ionization energy for metal  
(b) high electron affinity for non metal  
(c) high lattice energy  
(d) all of a-c above  
e) none of a-c above

5. Which of the atoms below is least likely to violate the octet rule?

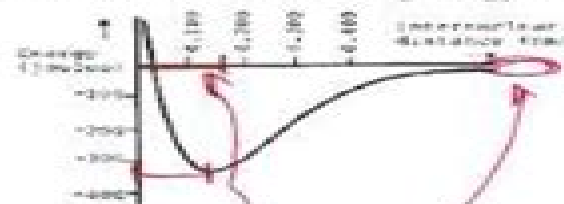
- a) Be b) P c) S d) B (e) F

6. How many electrons are shown in the Lewis structure of perchlorate ion,  $\text{ClO}_4^-$ ?

- a) 30 b) 31 (c) 32 d) 50 e) 51



- Questions 7 - 9 refer to the following energy diagram:



7. What is the energy of the two isolated atoms?

- a) -400 J d) 0.100 nm  
b) -335 J e) -155 J  
(c) 0 J

8. What is the bond length of the bond between the two atoms?

- a) 0.020 nm d) 0.330 nm  
(b) 0.140 nm e) 2.00 nm  
c) 0.400 nm

9. What is the bond energy (bond strength) of the bond?

- a) 0 J (d) -330 J Should this be + value? Unknown  
b) -110 J e) -422 J  
c) -10 J

10. What is the bond order of the C-O bond in acetone?



- a) 4 b) 1.5 c) 0.5 d) 1 (e) 2

# Geometry Chapter 2 Answers

**Scott A. Hatch, Lisa Zimmer Hatch**



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## **Table of Contents Geometry Chapter 2 Answers**

1. Understanding the eBook Geometry Chapter 2 Answers
  - The Rise of Digital Reading Geometry Chapter 2 Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Geometry Chapter 2 Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Geometry Chapter 2 Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Geometry Chapter 2 Answers
  - Personalized Recommendations
  - Geometry Chapter 2 Answers User Reviews and Ratings
  - Geometry Chapter 2 Answers and Bestseller Lists

5. Accessing Geometry Chapter 2 Answers Free and Paid eBooks
  - Geometry Chapter 2 Answers Public Domain eBooks
  - Geometry Chapter 2 Answers eBook Subscription Services
  - Geometry Chapter 2 Answers Budget-Friendly Options
6. Navigating Geometry Chapter 2 Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Geometry Chapter 2 Answers Compatibility with Devices
  - Geometry Chapter 2 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Geometry Chapter 2 Answers
  - Highlighting and Note-Taking Geometry Chapter 2 Answers
  - Interactive Elements Geometry Chapter 2 Answers
8. Staying Engaged with Geometry Chapter 2 Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Geometry Chapter 2 Answers
9. Balancing eBooks and Physical Books Geometry Chapter 2 Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Geometry Chapter 2 Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Geometry Chapter 2 Answers
  - Setting Reading Goals Geometry Chapter 2 Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Geometry Chapter 2 Answers
  - Fact-Checking eBook Content of Geometry Chapter 2 Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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