



Practical Sewing Advice

2

SEWING GUIDE

Dear User,

The Instruction Booklet shows you how to operate your Elna.

This Sewing Guide contains practical hints for easy sewing. It will help you to get the best out of your Elna and spend many happy hours with it.

Although far from being exhaustive, the Guide gives you a good idea of the wide range of possibilities for practical and decorative sewing that are available to you, either by using the standard equipment of your machine, or by acquiring one of the additional extra accessories.

Remember to inquire regularly at Elna shops and you will constantly make interesting discoveries. If you want to look up something in a hurry, consult the index on the last page.

Needle, Thread and Material

It is preferable to use good quality dull or mercerized thread. For darning, use fine left-twist thread, because it is stronger. Remember that a dry thread becomes brittle. The best sewing machine will give poor results with a bad needle.

A golden rule:

Select the thread according to the type of work and material, and the needle according to the thread.

Darning, sewing or embroidery thread, depending on the fineness, durability or appearance of the job to be sewn.

First, get used to simple sewing and you will be able to make use of other Elna possibilities with greater confidence, later on.

The type of thread or needle size will be referred to subsequently only in special cases.

Practical Sewing Advice

for the:

Elna Lotus-ec

Elna Lotus-az

Elna Lotus-sp

Elna Lotus-bp

Other sewing work and extra accessories

See pages:

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Materials	Threads			Needle Size System 150/755A
	Cotton	Silk	Synthetic Fibres	
Sheer	120-100 *	Sewing	Sewing	60-70 7-10
Wool (darning)	Darning 120-100 *	Darning	Darning	60-70 7-10
Ordinary	100-60 *	Sewing	Sewing	60 12
Medium Weight	60-50 *	Sewing	Sewing	60-80 12-14
Thick Woollen Heavy Shirts	40 *	Sewing	Sewing	80-110 14-18

* Size may change depending on the country ** On the system 205

■ All common fabrics can be sewn with no trouble at all. For certain man-made fibres, knitwear, stretch fabrics, silk, lingerie, jersey-wear, etc., it may be advisable to de-centre the needle to the left. The stitching will be even more regular.

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Elna Sp Manual Introduction

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company is in a strong position for the historical results from 2003 to 2007 the margin remains at the percentage between 15 and 20 for the five years however the gross margin rate with the project will also be strong

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web the attached document is a weekly lesson plan that includes a common core reading skill grammar spelling and vocabulary i use harcourt trophies as well as picture books to teach the reading skill i include kagan strategies global awareness and technology integrations i also include reading

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web although i mainly focus on 2nd grade guided math curriculum i wanted to do something a little different and share some exciting activities i used to do with my 1st 2nd grade students for johnny appleseed day this is such a fun topic for students and it seems like a seasonal theme makes everything more enjoyable

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web it is run by a fellow teacher and has materials for all of the stories for harcourt 2nd grade i also use a lot of her resources in my classroom johnny appleseed kid s page

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web the purpose of the copper iron stoichiometry lab was to determine which of the two possible iron sulfate compounds formed when reacting aqueous copper ii sulfate with solid iron metal the techniques used in this lab were weighing by difference quantitative transfer and vacuum filtration

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the iron metal dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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web lab report stoichiometry copper sulfate and iron the oxidation of trimethylamine by copper ii chloride nov 10 2022 this research was on the irreversible oxidation of trimethylamine by copper ii chloride which takes place at temperatures of about 75 and higher the coordination of trimethylamine by copper ii

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the

equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

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web under the finalise of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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