November 5,2020

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POI Ivan Lord T. Delos Reyes
2020 - 20091
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1. a acidic medium

i. As,
$$O_3$$
 (s) + NO_3 (aq) \longrightarrow H_3 As O_4 (aq) + N_2 O_3 (aq)

As, O_3 (s) \longrightarrow H_3 As O_4 (aq)

 $5H_2O + As_2O_3(c) \longrightarrow 2H_3$ As O_4 (aq) + $4H^{\dagger} + 4e^{-}$
 NO_3 (aq) \longrightarrow N_2O_3 (aq)

 $4e^{-} + (6H^{\dagger} + 2NO_3^{\dagger}(aq)) \longrightarrow N_2O_3 + 3H_2O$
 $2h_1^{\dagger}O + As_2O_3(c) \longrightarrow 2H_3$ As O_4 (aq) + $9H^{\dagger} + 4e^{-}$
 $Ae^{+} + (4H^{\dagger} + 2NO_3^{\dagger}(aq)) \longrightarrow N_2O_3 + 3H_2O$

 $2H_{2}O + AS_{2}O_{3}(S) + 2H^{\dagger} + 2NO_{3}^{-} \rightarrow 2H_{3}ASO_{4}(aq) + N_{2}O_{3}$ ii 1. (s) + OCI (aq) \rightarrow IO (aq) + CI (aq)

$$\begin{array}{c} 1_{2}(s) + 0Cl(aq) \longrightarrow IO_{3}(aq) + Cl(aq) \\ & \pm_{2}(s) \longrightarrow IO_{3}(aq) \\ & (4H_{2}O + I_{2}(s) \longrightarrow 2IO_{3}(aq) + 12H' + 10e' \\ & 0Cl(aq) \longrightarrow Cl(aq) \\ & (2e' + 2H' + 0Cl(aq) \longrightarrow Cl' + H_{2}O) / 5 \\ & 10e' + 10H' + 5Ocl(aq) \longrightarrow 5cl' + 5H_{2}O, \\ & (4H_{2}O + I_{2}(s) \longrightarrow 2IO_{3}(aq) + 12H' + 10e' \\ & (9e' + 10H + 5Ocl(aq) \longrightarrow 5cl' + 5H_{2}O) \\ & (4H_{2}O + I_{2}(s) \longrightarrow 5cl' + 5H_{2}O) \\ & (4H_{2}O + I_{2}(s) \longrightarrow 5cl' + 5H_{2}O) \end{array}$$

Electrochemistry Practice Problems

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