MATTER AND CHANGE

SECTION 2.1 PROPERTIES OF MATTER (pages 39-42)

This section helps you distinguish extensive from intensive properties and identify substances by their properties. It teaches you how to differentiate the three states of matter. It also defines a physical property and lists examples of pleysical properties and physical charges.

▶ Describing Matter (page 39)

- of an object is a measure of the amount of matter the object contains.
- 2. How does an extensive property differ from an intensive property?

An extensive property depends on the amount of matter, while an intensive property depends on the type of matter.

without

▶ Identifying Substances (page 40)

3. Matter that has a uniform and definite composition is called a

- 4. Is the following sentence true or false? All samples of a substance have Table 1 different physical properties.
- 5. A physical property is a quality or condition of a substance that can be measured

changing the substance's composition.

6. Circle the letter of the term that is NOT a physical property.

COLD

- a. bandos ss
- e. besilving posicit
- by contain
- d. m elting
- 7. Look at Table 2.1 on page 40. What is the melting point of bromine? _______
- 8. Look at Table 2.1 on page 40. Circle the letter of the substance that is a yellow solid and melts at 115°C.

on exhalthour

- b. chlorine
- c. gold:
- d. copper

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Victor M. Corman

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