Name of the student

Date – 14.09.2022 (Wednesday) Section

GENERAL INSTRUCTIONS:

- All questions are compulsory.
- Question nos. 1 to 21 are MCQs with one correct option and carrying 1 mark each.
- Question nos. 22 to 24 are objective/short answer type questions.
- Question nos. 25 to 28 are of 2 marks each.
- Question nos. 29 to 32 is of 3 marks each.
- Question nos. 33 to 35 is of 5 marks each.

SECTION - A

Q. 1 to 21 is of 1 mark each.

- Q.1 Which of the following aqueous solution should have highest boiling point?
 - a) 1 M NaOH
- b) 1 M Na₂SO₄
- c) 1M NH₄NO₃
- d) 1 M KNO₃

- Q.2 Which of the following statement is false?
 - a) Two different solution of sucrose of same molality prepare in different solvent will have the same depression in freezing point.
 - b) The osmotic pressure of a solution is given by the equation π =CRT. (where C is the molarity of the solution)
 - c) Decreasing order of osmotic pressure for 0.01M aqueous solution of Barium Chloride, Potassium chloride, acetic acid and sucrose is BaCl₂ > KCl > CH₃COOH > Sucrose
 - d) According to Raoult's Law, the vapour pressure exeterd by a volatile component of solution is directly proportional to its molar fraction in the solution.
- Q.3 We have three aqueous solution of NaCl labelled as 'A', 'B' and 'C' with concentration 0.1 M, 0.01 M and 0.001 M respectively. The value of Van't Hoff factor for these solution will be in the order
 - a) ia < in < ic
- a) $i_A > i_B > i_C$
- a) $i_A = i_B = i_C$

- a) iA < iB > ic
- Q.4 On the basis of the information given below mark the correct options -
 - In bromoethane and chloroethane mixture intermolecular interaction of A-A and B-B type are nearly same as A-B type interaction.
 - ii) In the ethanol and acetone mixture A-A or B-B type intermolecular interactions are stronger than A-B type interaction.

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with a view to its relevance to other institutions. The major part of the report consists of findings from field observations of central administration selection committees and faculty office procedures as selection decisions were actually made over two years. Issues and problems are identified and discussed in detail in relation to general policy and administrative considerations. A brief review and technical appendix includes information of a statistical nature on the prediction of success in university courses. Strategies for change are discussed in relation to a general review of admissions policy including the introduction of a special admissions scheme and the possible use of achievement tests. University policy documents are included in the appendix Although some of the problems at the University of Melbourne relate to the intense competition for entry to this particular university many of the issues addressed and solutions proposed deal with common problems and will suggest policy options and procedures of interest to other universities and colleges and to individuals concerned with questions of educational quality and social equity.

Excel HSC Physics Neville G. Warren, 2003

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