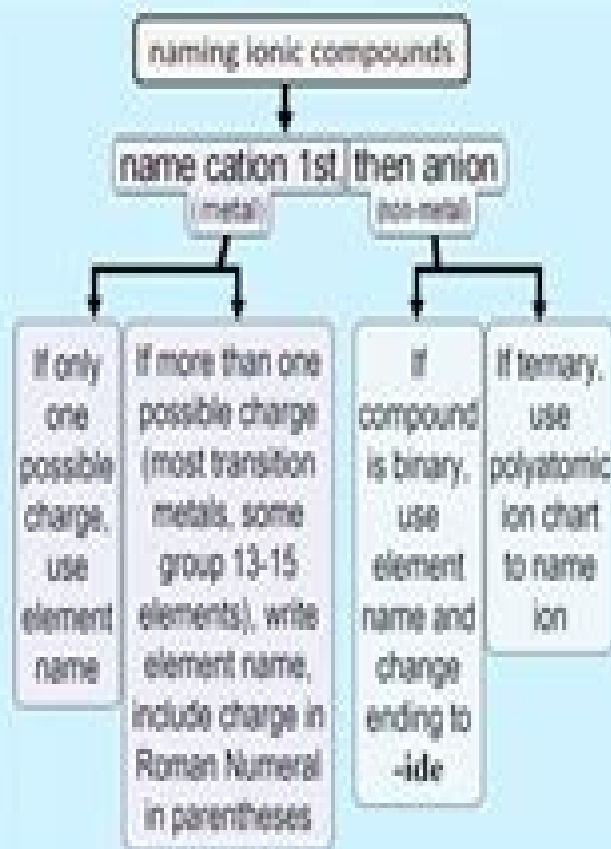


# Naming Ionic Compounds



Try the following:

FORMULA	NAME
$\text{SrCl}_2$	strontium chloride
$\text{CoCl}_2$	cobalt (II) chloride
$\text{Ba}_3(\text{AsO}_4)_2$	barium arsenate
$\text{Pt}_3(\text{AsO}_4)_2$	platinum (II) arsenate
$\text{Pt}_3(\text{AsO}_4)_4$	platinum (IV) arsenate
$\text{Cu}_3\text{N}$	copper (I) nitride
$\text{CuSO}_3$	copper (II) sulfite

— *Crash*Chemistry —

# Naming Molecular Compounds Study Guide

**LL Leslie**

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