Name:	KEY	Block:
The state of the s		

11. MOBr. Malybadenium (II) Bramide

12. VCI Vanadium (V) Chlaride

13. Mn2O3 Manganese (III) Oxide

15. NP2O3 NEDTUNIUM (II) OXICE

16. V2Os Vanadium (V) Oxide

14. COO _ COLONH (II) Oxide

Assignment #2 - Compound Names and Formulas Multi-valent ions only

1. copper (II) oxide	11. manganese (III) oxide Mr 203
2. mercury (I) oxide +\0,20	12. vanadium (II) bromide VB(z
B. gold (III) chloride _AuCl3	13. niobium(V) oxide
4. thallium (III) bromide TIPY 3	14. titanium (III) oxide T\20>
5. bismuth (V) oxide <u>B1205</u>	15. titanium (III) nitride TIN
3. terbium (III) oxide	16. iron (II) oxide FeO
. uranium (VI) oxide \(\lambda \)	17. cobalt (II) phosphide C03P2
protactinium (IV) bromide PhBY4	18. tin (II) oxide
cerium (III) oxide	19. thulium (II) bromide _\TWBY ≈
10. arsenic (V) sulphideA52S5	20. copper (II) bromide CUBY2
Name the compound using the Roman Nume	eral Method and balance the ionic charges.
. snc4 Tin (IV) Chloride	9. PdF. Palladium (IV) Fluoride
2. BIBIS _ BISMINTO (V) Bromide	10. Osoo Osmium (III) Oxide

3. POO2 POLONIUM (IV) OXICE

6. HgCI <u>HEYCUYY</u> (I) Chloride

5. Hgo Mercury (II) Oxide

4. Pblz _ I cod(I) Iodide

7. Auzos Gold (III) Oxide.

8. FeCI Iron (II) Chlorida

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M Walker

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