

Three Hydrogen-like Atoms – Simultaneous Quantities

Electromagnetic Radiation

Consider now the hydrogen atom, which emits energy levels, light, and transitions and also is the following quantities in terms of its understanding:

Quantum is a concept that describes the behavior of particles, such as the light, which is the electromagnetic radiation. The hydrogen atom, however, can be thought of as a system of particles, such as the electron, which is the electron.

Quantity	Value (Hydrogen)	Quantum Number
Energy	Electron's energy level in the atom	Electron's energy level in the atom
Electron	Electron's energy level in the atom	Electron's energy level in the atom
Photon	Electron's energy level in the atom	Electron's energy level in the atom
Electron's energy level in the atom	Electron's energy level in the atom	Electron's energy level in the atom
Energy of the atom or photon	$E_{\text{atom}} = E_{\text{photon}} + E_{\text{electron}}$	$E_{\text{atom}} = E_{\text{photon}} + E_{\text{electron}}$

Quantum is a concept that describes the behavior of particles, such as the light, which is the electromagnetic radiation. The hydrogen atom, however, can be thought of as a system of particles, such as the electron, which is the electron.

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