

(d)	$\text{C}_5\text{H}_{12}(\text{g}) + 3\text{O}_2(\text{g}) \rightarrow 5\text{C}(\text{s}) + 6\text{H}_2\text{O}(\text{l})$ <p>OR</p> $\text{C}_5\text{H}_{12}(\text{g}) + \frac{11}{2}\text{O}_2(\text{g}) \rightarrow 5\text{CO}(\text{g}) + 6\text{H}_2\text{O}(\text{l})$	(1 mark)
(e)	- Decomposition / breakdown of organic matter ✓ 1 in the absence of oxygen. ✓ 1	(2 marks)
(f)	- Heat a mixture of sodium ethanoate and sodalime ✓ 1 - methane gas is produced and collected over water/ in a syringe/upward delivery ✓ 1	(2 marks)
		<b>13 marks</b>

2.	<p>(a) (i) - Dynamic equilibrium refers to a state of balance/reaction in which forward and backward/reverse reactions are taking place at the same rate. ✓ 1</p> <p>(ii) - The intensity of the orange colour increases. ✓ 1</p> <p>Addition of <math>\text{H}^+</math> ions, sulphuric(VI) acid removes <math>\text{OH}^-</math> ions from the equilibrium mixture hence equilibrium shifts to the left. ✓ 1</p> <p>(b) (i) Nitrous acid / <math>\text{HNO}_2</math>/ Nitric(III) acid</p>	<p>(1 mark)</p> <p>(2 marks)</p> <p>(1 mark)</p>
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# Kcse Chemistry Paper 20with Answers

**Marcel A. Müller**



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