

## The Hydrogen Atom – Student Guide

### Background Material

Carefully read the background pages created through Eureka, Slides, and Experiments and answer the following questions in detail and with understanding.

**Question 1:** Complete the following table which compares how the Bohr model and the Quantum model compare the Hydrogen atom. In each case, they both describe things in a unique way and in some ways they do not.

| Concept<br>Name                                | Bohr Model                               | Quantum Model                                                      |
|------------------------------------------------|------------------------------------------|--------------------------------------------------------------------|
| Electrons                                      | positively charged particles in the atom | positively charged particles in the atom                           |
| Orbitals                                       | negatively charged and orbit the nucleus | negatively charged clouds                                          |
| Locations of the electrons in the atom without | electrons have fixed positions           | electrons cloud particles<br>occupies the space around the nucleus |
| Energy of the electron levels                  | $E = E_n$                                | $E = E_n$                                                          |

**Question 2:** Complete the following table which shows the differences of two different theories by putting predictions around the comparison statements. The table was completed for you to help illustrate a unique strength, another hypothesis, another strength, and the same reality through science in the future.

# Hydrogen Atom Student Guide Naap Answers

**M Walker**



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