

Name: \_\_\_\_\_

## Honors Chem Test: Ch. 7

**Multiple Choice—1 point each**

Identify the choice that best completes the statement or answers the question.

- How many atoms of fluorine are present in a molecule of carbon tetrafluoride,  $\text{CF}_4$ ?
  - 1
  - 2
  - 4
  - 5
- Name the compound  $\text{Al}_2\text{S}_3$ .
  - aluminum sulfate
  - aluminum sulfur
  - aluminum(II) sulfate
  - aluminum sulfide
- Name the compound  $\text{CF}_4$ .
  - calcium fluoride
  - carbon fluoride
  - carbon tetrafluoride
  - mono-carbon quadrafluoride
- Name the compound  $\text{N}_2\text{O}_5$ .
  - dinitrogen oxide
  - nitrogen trioxide
  - nitric oxide
  - dinitrogen trioxide
- What is the formula for diarsenic trisulfide?
  - $\text{As}_2\text{S}_3$
  - $\text{AsS}_3$
  - $\text{As}_3\text{S}_4$
  - $\text{As}_3\text{S}_3$
- What is the formula for sulfur dichloride?
  - $\text{SCl}$
  - $\text{SCl}_2$
  - $\text{S}_2\text{Cl}$
  - $\text{S}_2\text{Cl}_2$
- What is the formula for hydrochloric acid?
  - $\text{HF}$
  - $\text{HCl}$
  - $\text{HClO}$
  - $\text{H}_2\text{CO}_3$
- The molar mass of an element is the mass of one
  - atom of the element.
  - crystal of the element.
  - gram of the element.
  - mole of the element.
- The molar mass of  $\text{MgI}_2$  is
  - the sum of the masses of 1 mol of Mg and 2 mol of I.
  - the sum of the masses of 1 mol of Mg and 1 mol of I.
  - the sum of the masses of 1 atom of Mg and 2 atoms of I.
  - the sum of the masses of 1 atom of Mg and 1 atom of I.

# Honors Chemistry Chapter Practice Test

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