

# SECTION - A

Q. 1. Select and write the correct answer for the following multiple choice type of questions :

[10]

(i) The product obtained in the following reaction

P 239  
11.2



- (a) Pent-3-en-1-ol (b) ☒ Pentan-1-ol  
(c) Pentan-2-ol (d) Pentanal

(ii) Amongst the following, the solubility of which ionic solid decreases with increase in temperature?

P 30  
11)

- (a)  $\text{KNO}_3$  (b)  $\text{NaBr}$   
(c) ☒  $\text{Na}_2\text{SO}_4$  (d)  $\text{KCl}$

(iii) The correct IUPAC name of  $\text{Na}_3 [\text{AlF}_6]$  is

P 196  
Table  
9.3

- (a) ☒ Sodium hexafluoroaluminate (III)  
(b) Sodium hexafluoroaluminate (II)  
(c) Sodium hexafluoroaluminium (III)  
(d) Sodium hexafluoroaluminium (II)

(iv) Which of the following acids has highest pKa value?

P 277  
Table  
12.9

- (a) Mono chloroacetic acid (b) Dichloroacetic acid  
(c) Trichloroacetic acid (d) ☒ Acetic acid

(v) Number of carbon atoms present in isoprene unit is \_\_\_\_.

P 322  
15.3

- (a) 6 (b) ☒ 5  
(c) 4 (d) 3

(vi) The colourless transition metal ion amongst the following is

P 173  
(a) ☒  $\text{Cu}^+$  3d<sup>10</sup>  
(c)  $\text{Ni}^{2+}$

- (b)  $\text{Cu}^{2+}$   
(d)  $\text{Co}^{2+}$

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1. Understanding the eBook Hsc Unregular Chemistry Question Paper 2bd
  - The Rise of Digital Reading Hsc Unregular Chemistry Question Paper 2bd
  - Advantages of eBooks Over Traditional Books
2. Identifying Hsc Unregular Chemistry Question Paper 2bd
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hsc Unregular Chemistry Question Paper 2bd
  - User-Friendly Interface
4. Exploring eBook Recommendations from Hsc Unregular Chemistry Question Paper 2bd
  - Personalized Recommendations
  - Hsc Unregular Chemistry Question Paper 2bd User Reviews and Ratings
  - Hsc Unregular Chemistry Question Paper 2bd and Bestseller Lists

5. Accessing Hsc Unregular Chemistry Question Paper 2bd Free and Paid eBooks
  - Hsc Unregular Chemistry Question Paper 2bd Public Domain eBooks
  - Hsc Unregular Chemistry Question Paper 2bd eBook Subscription Services
  - Hsc Unregular Chemistry Question Paper 2bd Budget-Friendly Options
6. Navigating Hsc Unregular Chemistry Question Paper 2bd eBook Formats
  - ePub, PDF, MOBI, and More
  - Hsc Unregular Chemistry Question Paper 2bd Compatibility with Devices
  - Hsc Unregular Chemistry Question Paper 2bd Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Hsc Unregular Chemistry Question Paper 2bd
  - Highlighting and Note-Taking Hsc Unregular Chemistry Question Paper 2bd
  - Interactive Elements Hsc Unregular Chemistry Question Paper 2bd
8. Staying Engaged with Hsc Unregular Chemistry Question Paper 2bd
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Hsc Unregular Chemistry Question Paper 2bd
9. Balancing eBooks and Physical Books Hsc Unregular Chemistry Question Paper 2bd
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Hsc Unregular Chemistry Question Paper 2bd
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Hsc Unregular Chemistry Question Paper 2bd
  - Setting Reading Goals Hsc Unregular Chemistry Question Paper 2bd
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Hsc Unregular Chemistry Question Paper 2bd
  - Fact-Checking eBook Content of Hsc Unregular Chemistry Question Paper 2bd
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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