

# Acids, Bases & pH Review - Honors Chemistry

Directions: For all multiple choice questions, highlight the correct answer(s). Answer short answer questions in the box provided.

## Part 1: Properties & definition of Acids & Bases

1. Complete the table below.

	ACIDS	BASES
Taste		
pH range		
Conducting electricity		
Neutralize		
Litmus		
Phenolphthalein		

2. An indicator

- is a type of acid.
- is a type of base.
- has a neutral pH.
- changes color depending on the pH.

3. Litmus and phenolphthalein are indicators. What does that mean?

4. Acids and bases "dissociate" in water to form  $H^+$  (for acids) and  $OH^-$  (for bases). What does "dissociate" mean in this context?

5. Identify whether each substance is an acid or a base.

- $HBr$  :
- $HCO_2H$  :
- $Al(OH)_3$  :
- $NH_3 + H_2O \rightarrow NH_4^+ + OH^-$ ;  $NH_3$  is a(n) :
- $H_2SO_4 + H_2O \rightarrow H_3O^+ + HSO_4^-$  ;  $H_2SO_4$  is a(n) :
- $HNO_3 \rightarrow H^+ + NO_3^-$  ;  $HNO_3$  is a(n) :

# Heath Chemistry Learning Guide Acids And Bases

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