

Experiment 10: Electrodes with a Standard Solution

From: *Basic Laboratory Chemistry Experiments*, pp. 45-46

The purpose of this experiment is to determine the standard electrode potential of some metal ions. In this experiment, a standard solution of a metal ion is used as a reference electrode. The standard electrode potential of the metal ion is determined by measuring the potential of the cell formed by the metal ion and a standard solution of the same metal ion. The standard electrode potential of the metal ion is determined by measuring the potential of the cell formed by the metal ion and a standard solution of the same metal ion. The standard electrode potential of the metal ion is determined by measuring the potential of the cell formed by the metal ion and a standard solution of the same metal ion.

In this experiment, you will determine the standard electrode potential of some metal ions. The standard electrode potential of the metal ion is determined by measuring the potential of the cell formed by the metal ion and a standard solution of the same metal ion.

Objectives:

- to determine the standard electrode potential of some metal ions
- to determine the standard potential of the standard solution of the metal ion
- to determine the standard potential of the metal ion
- to determine the standard potential of the standard solution of the metal ion
- to determine the standard potential of the standard solution of the metal ion

Materials:

- standard solution
- standard solution (100 mL)
- standard solution
- standard solution
- standard solution
- standard solution
- standard solution
- standard solution
- standard solution

Procedure:

- to determine the standard electrode potential of some metal ions
- to determine the standard potential of the standard solution of the metal ion
- to determine the standard potential of the metal ion
- to determine the standard potential of the standard solution of the metal ion
- to determine the standard potential of the standard solution of the metal ion

Results:

- to determine the standard electrode potential of some metal ions
- to determine the standard potential of the standard solution of the metal ion
- to determine the standard potential of the metal ion
- to determine the standard potential of the standard solution of the metal ion
- to determine the standard potential of the standard solution of the metal ion

Discussion:

- to determine the standard electrode potential of some metal ions
- to determine the standard potential of the standard solution of the metal ion
- to determine the standard potential of the metal ion
- to determine the standard potential of the standard solution of the metal ion
- to determine the standard potential of the standard solution of the metal ion

Heath Chemistry 1lab 6b Answers

Daniela Niemeyer



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