

### Mole/Mole Ratio Problems



- How many moles of hydrogen are needed to completely react with two moles of nitrogen?
- How many moles of nitrogen trihydride can be produced with 5 moles of nitrogen?
- How many moles of nitrogen are needed to produce .5 moles of nitrogen trihydride?
- How many liters of nitrogen trihydride can be produced from 24 liters of nitrogen gas?



- How many moles of oxygen are produced by the decomposition of six moles of potassium chlorate?
- How many moles of potassium chloride are produced by the decomposition of .75 moles of potassium chlorate?
- How many moles of potassium chlorate are needed to produce 100 moles of oxygen?
- How many liters of oxygen will the decomposition of 100 liters of potassium chlorate produce?

# Mole Ratio Worksheet Model 1

**AN Whitehead**



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