

Name: _____ Date: _____ Period: _____

NOVA: Hunting the Elements

Part 1: Basic Chemistry

Blank Periodic Table

Fill in the parts of this periodic table covered by the documentary.

1 H 1.00794																	2 He 4.002602	
3 Li 6.941	4 Be 9.00918											5 B 10.811	6 C 12.011	7 N 14.00643	8 O 15.999	9 F 18.998403	10 Ne 20.1797	
11 Na 22.98976928	12 Mg 24.304											13 Al 26.9815386	14 Si 28.08558	15 P 30.973762	16 S 32.06	17 Cl 35.453	18 Ar 39.948	
19 K 39.0983	20 Ca 40.078	21 Sc 44.955912	22 Ti 47.88	23 V 50.9415	24 Cr 51.9961	25 Mn 54.938045	26 Fe 55.845	27 Co 58.933195	28 Ni 58.6934	29 Cu 63.546	30 Zn 65.38	31 Ga 69.723	32 Ge 72.64	33 As 74.9216	34 Se 78.96	35 Br 79.904	36 Kr 83.80	
37 Rb 85.4678	38 Sr 87.62	39 Y 88.90584	40 Zr 91.224	41 Nb 92.90638	42 Mo 95.94	43 Tc 98.9062	44 Ru 101.07	45 Rh 102.9055	46 Pd 106.90558	47 Ag 107.8642	48 Cd 112.411	49 In 114.818	50 Sn 118.710	51 Sb 121.757	52 Te 127.6	53 I 126.90548	54 Xe 131.29	
55 Cs 132.90545196	56 Ba 137.327	57-70 (*)	71 Lu 174.967	72 Hf 178.49	73 Ta 180.94788	74 W 183.84	75 Re 186.207	76 Os 190.23	77 Ir 192.222	78 Pt 195.084	79 Au 196.966569	80 Hg 200.59	81 Tl 204.3833	82 Pb 207.2	83 Bi 208.980399	84 Po 209	85 At 210	86 Rn 222
87 Fr 223	88 Ra 226	89-102 (*) (*)	103 Lr 260	104 Rf 261	105 Db 262	106 Sg 266	107 Bh 264	108 Hs 277	109 Mt 268	110 Uun 289	111 Uuu 289	112 Uub 289	114 Uuq 289					

^a Lanthanide series.

57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb
89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No

[illegible]

Gold - Au

1. Write the number of subatomic particles in gold:
 - a. Protons;
 - b. Neutrons;
 - c. Electrons;
2. Give one property of gold.
3. How much gold is in one ton of the mined rock?

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